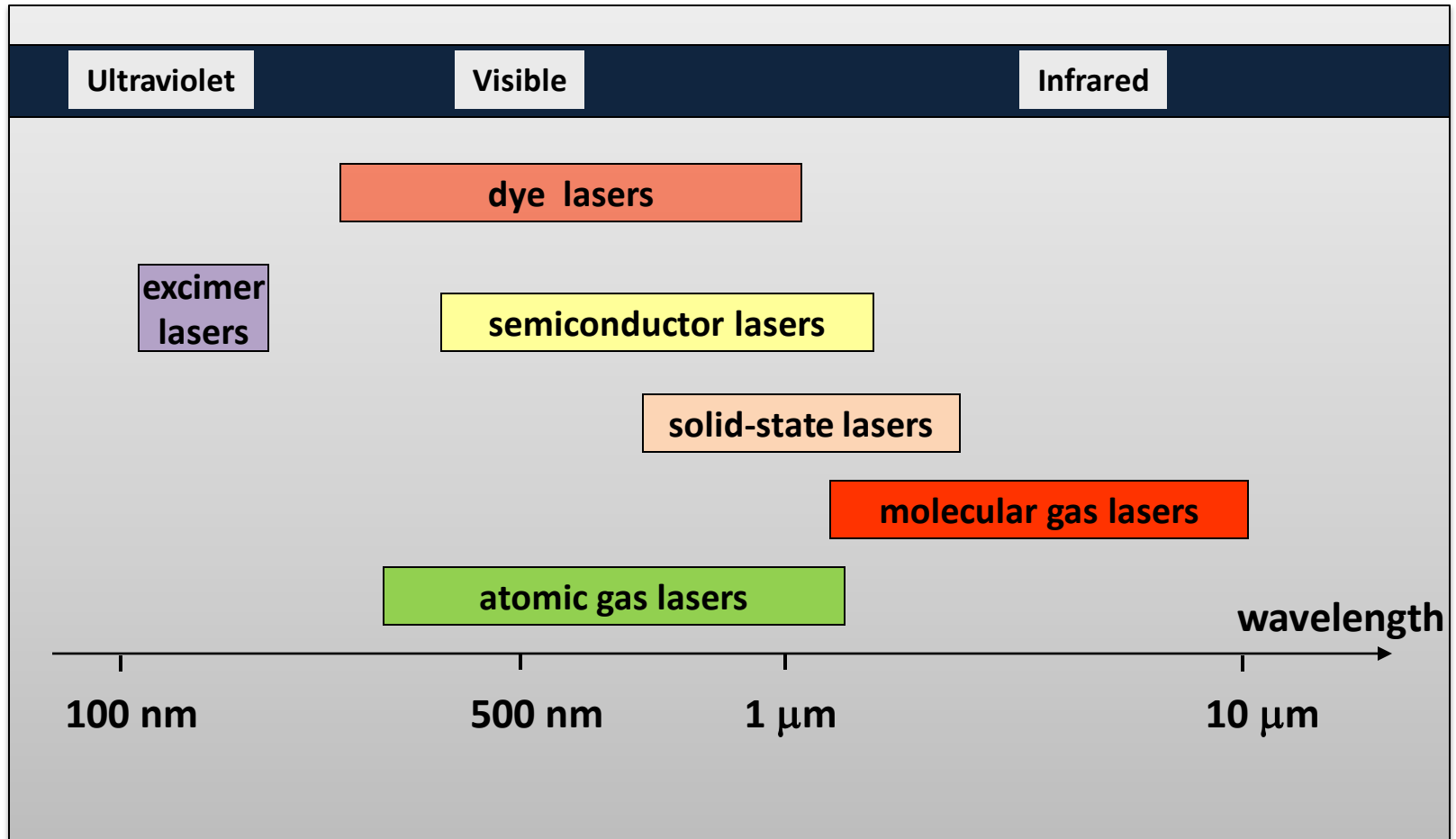


Examples of Specific Laser Systems

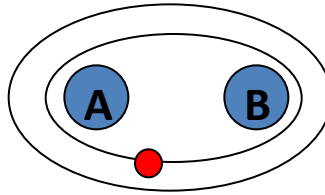
- Gas Lasers
CO₂ 200+ kW
- Solid-State Lasers
Nd:YAG (15 kW)
- Fiber Lasers
Yb³⁺ (5+ kW)
- Dye Lasers
- Chemical Lasers
COIL (7+kW), MIRACL (>1 MW !!)
- Semiconductor Lasers

6.5 Active media and spectral ranges



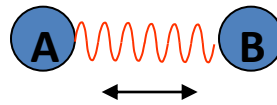
emission

electronic transitions



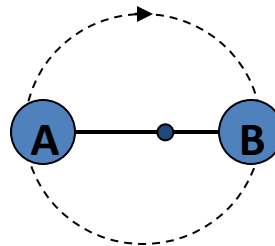
VIS, UV

vibrational transitions

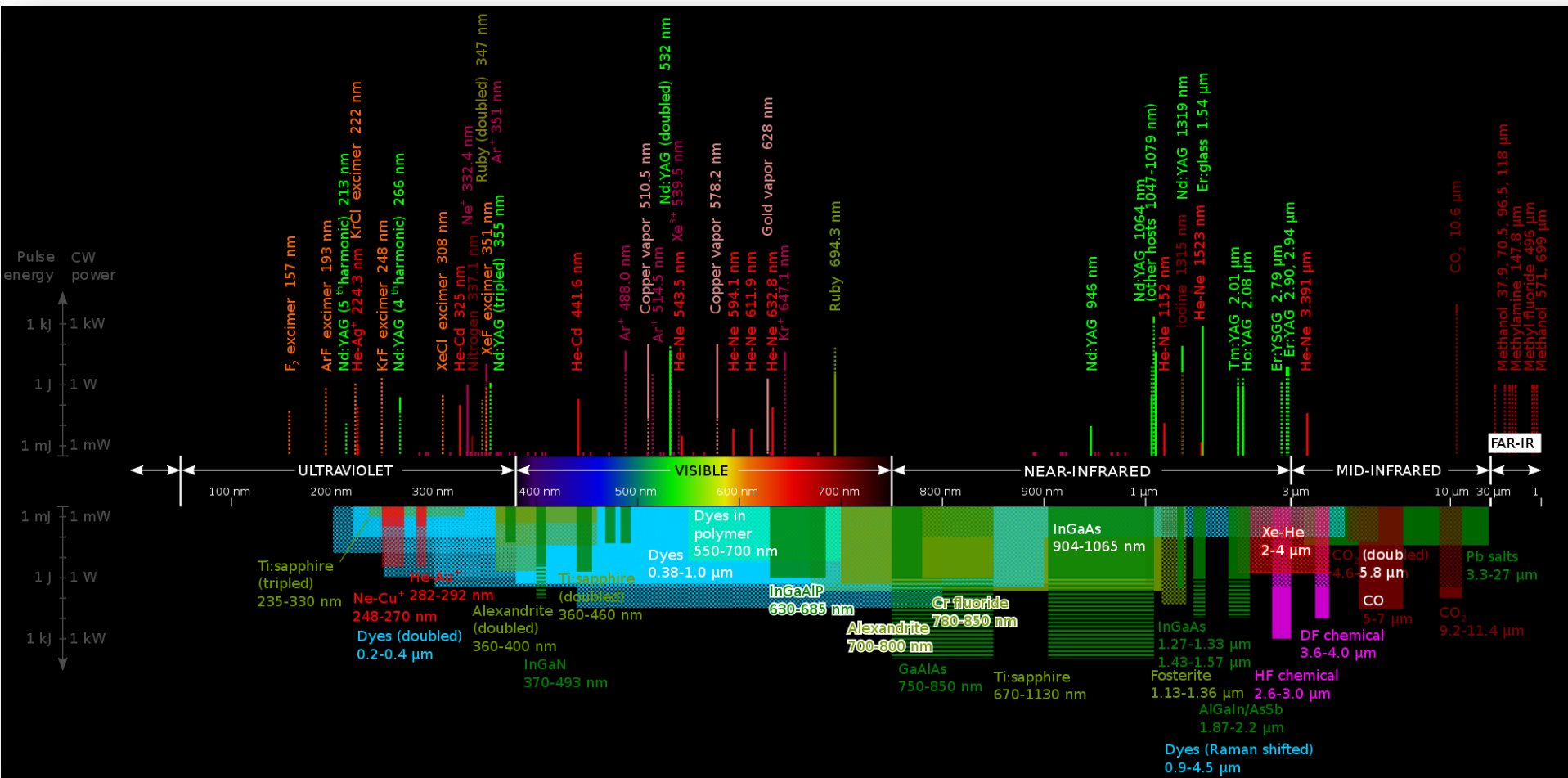


NIR, IR

rotational transitions



FIR

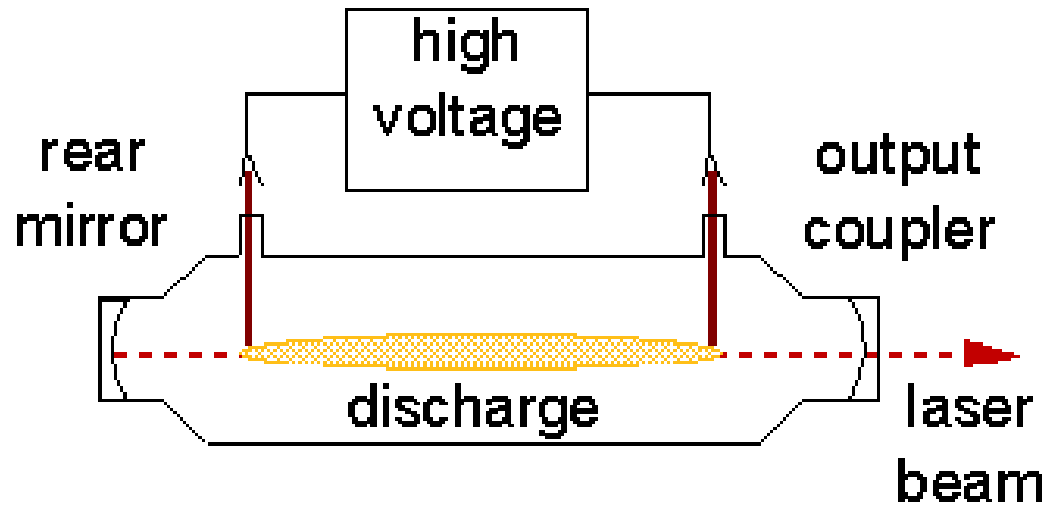


Typical laser efficiencies η :

$$\eta = \frac{\text{output power}}{\text{electrical input power}}$$

| | |
|-----------------------|--------|
| Argon - ion | < 0.1% |
| CO ₂ laser | < 20% |
| Excimer | < 20% |
| GaAlAs (diode laser) | < 40% |
| HeNe | < 0.1% |
| Nd:YAG | < 10% |

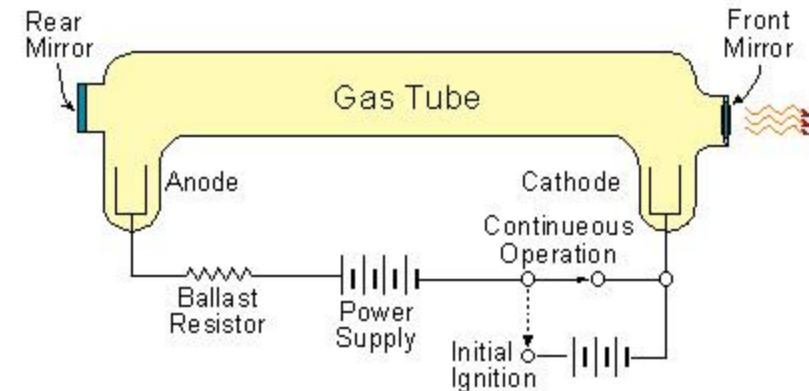
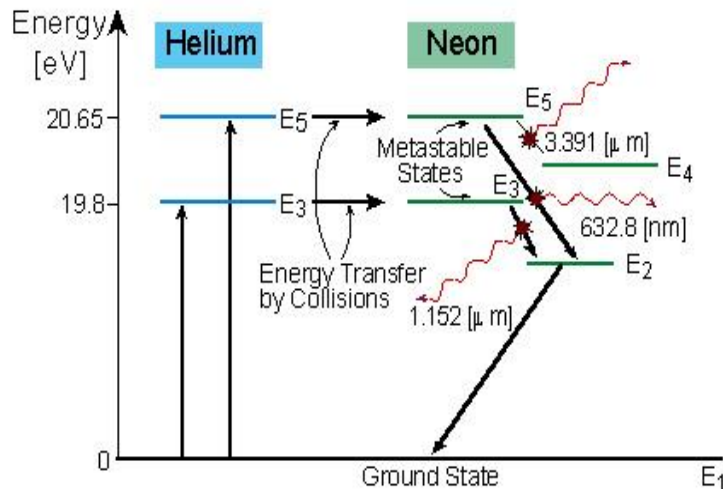
Gas Lasers



The excitation mechanism in most gas lasers is via *electric discharge*

The first Gas Laser: He-Ne

Ali Javan, et al. (Bell Labs, 1962)



- The second working LASER system to be demonstrated.
- The first gas LASER to be produced.
- The first LASER to produce a continuous output beam
- The active laser medium is a gaseous mixture of He & Ne atoms, in a roughly 10:1 proportion
- The gas is enclosed in a cylindrical quartz DISCHARGE tube

Comparison of Gas Lasers

| <i>Laser Type</i> | <i>Linear Power Density W/m</i> | <i>Maximum Power W</i> | <i>Power Efficiency percent</i> |
|--------------------------|--|-----------------------------------|--|
| He-Ne | 0.1 | 1 | 0.1 |
| Argon | 1-10 | 50 | 0.1 |
| CO ₂ | 60-80 | >10 ⁴ | 15-20 |

C. K. N. Patel, "Continuous-Wave Laser Action on Vibrational Rotational Transitions of CO₂," *Physics Review*, Vol. 136 A, (Nov., 1964) P. 1187

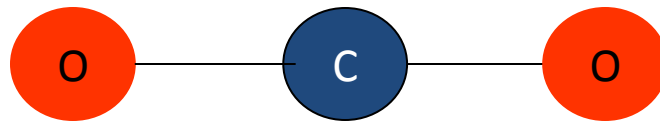


Applications (*peeling peanuts to star wars*)

- Industrial (cutting, welding, material processing)
- Military (range finding, targeting, remote sensing, sensor blinding, destroying ...)
- Medical (cutting, skin resurfacing)
-

11.2 Molecular Vibrations and Rotations

- Transitions are between molecular vibrational-rotational levels.

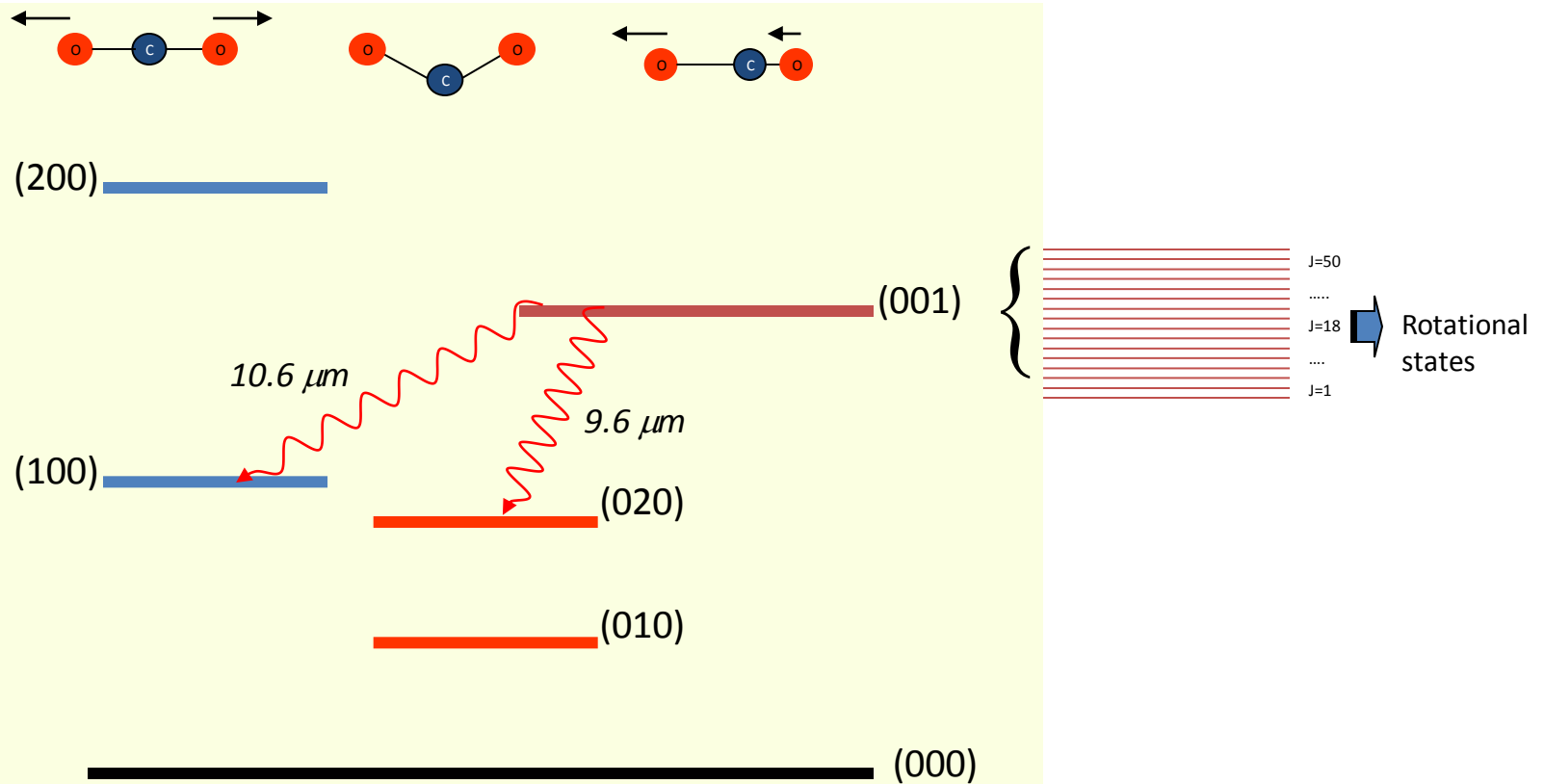


Modes of vibrations:

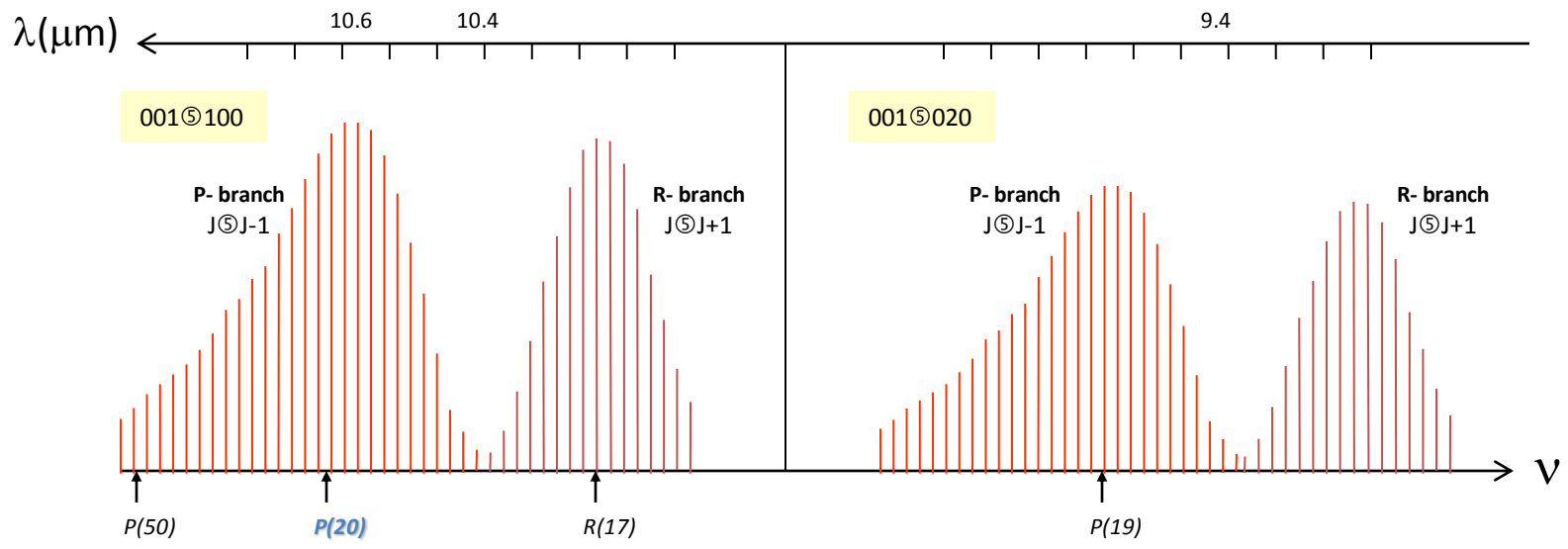
- Symmetric stretch
- Asymmetric stretch
- Bending mode

Simple Harmonic Oscillator (Quantum Mechanics):

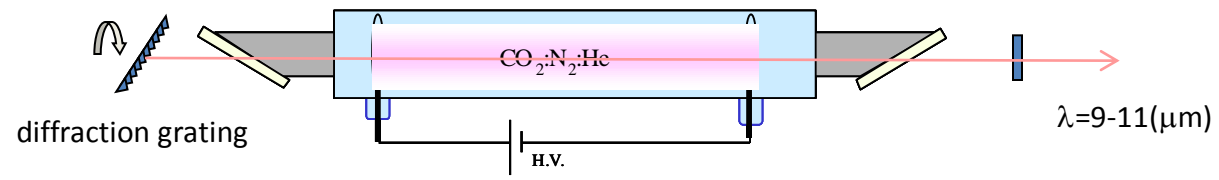
$$E(n_1, n_2, n_3) = h\nu_1(n_1 + 1/2) + h\nu_2(n_2 + 1/2) + h\nu_3(n_3 + 1/2)$$

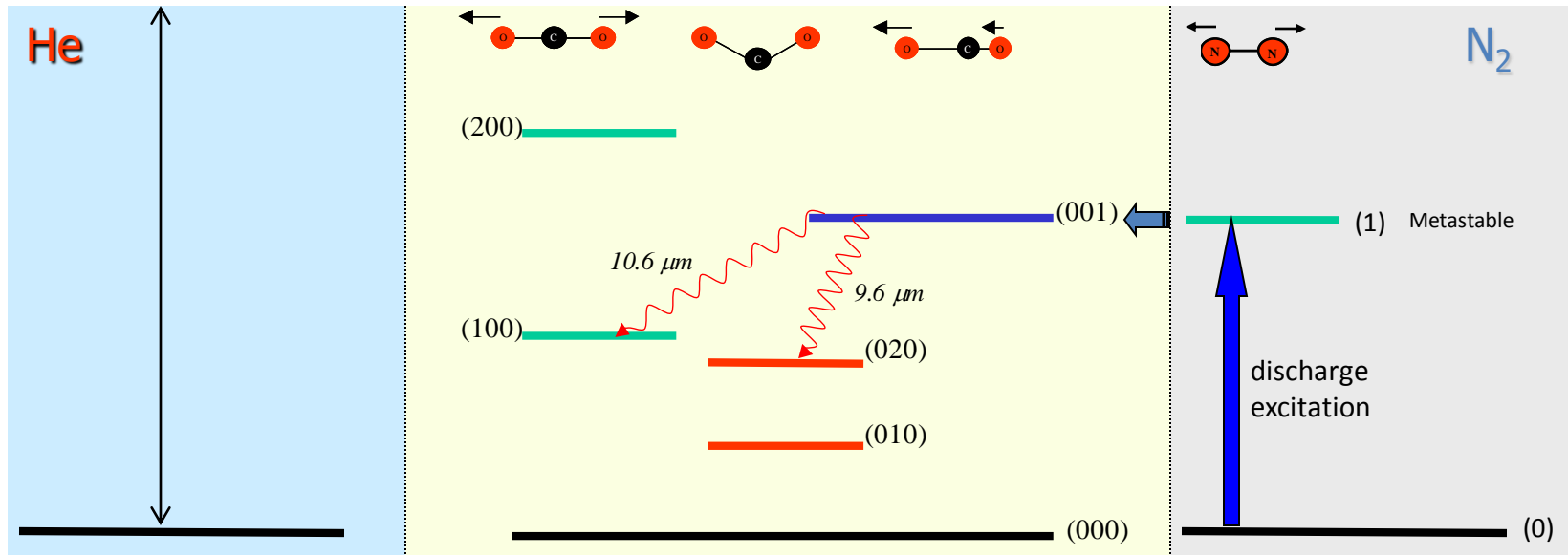


CO₂ Laser Transitions



Tuning:



Effect of Gas Mixtures: $\text{CO}_2 + \text{N}_2 + \text{He}$ 

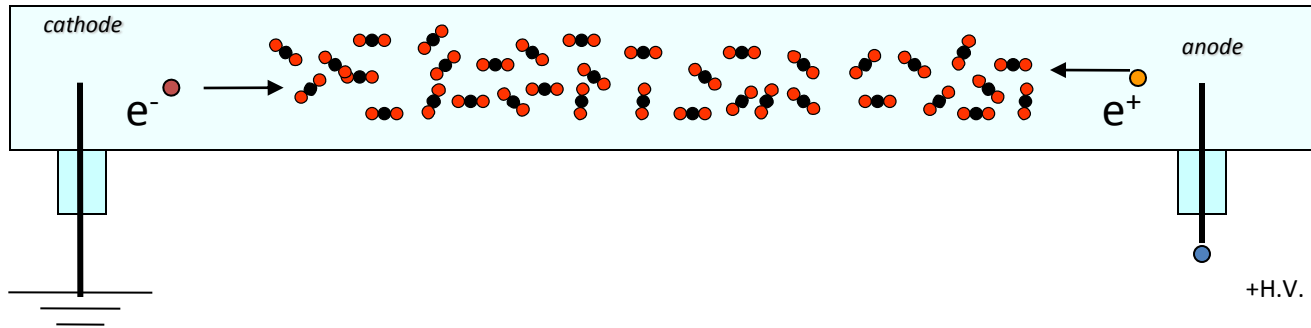
- Nitrogen helps populating the upper laser level in a discharge
- Helium helps to depopulate the lower laser level by collisions

Other possible additions to the gas mixture: CO, H₂

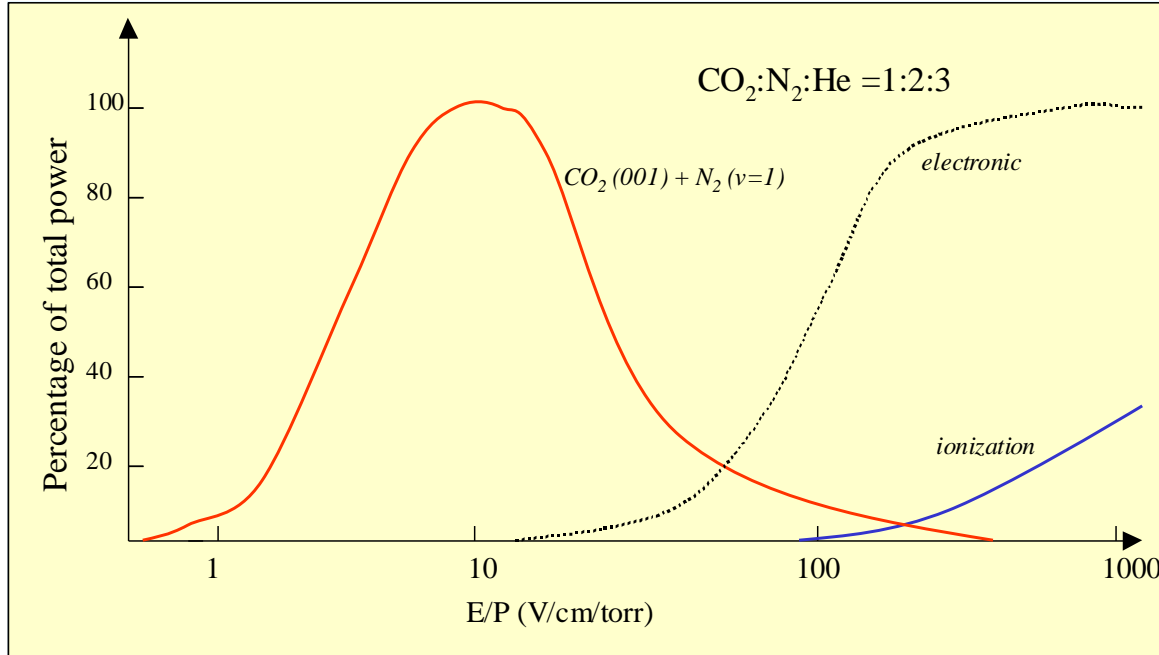
Typical CO₂:N₂:He Gas Ratios Recommended by Laser Manufacturers

| CO₂ | N₂ | He | Laser Power Rating W |
|-----------------------|----------------------|-------------|-----------------------------|
| 1 | 3 | 17 | 20 |
| 1 | 1.5 | 9.3 | 50 |
| 1 | 1.5 | 9.3 | 100 |
| 1 | 1.35 | 12.5 | 275 |
| 1 | 8 | 23 | 375 |
| 1 | 6.7 | 30 | 525 |
| 1 | 2.3 | 17 | 1000 |

11.3 Gas Discharge Phenomena



- Electrons emitted from cathode get accelerated by the electric field
- The energetic electrons excite the vibrational modes of the gas molecule via inelastic collisions



Example:

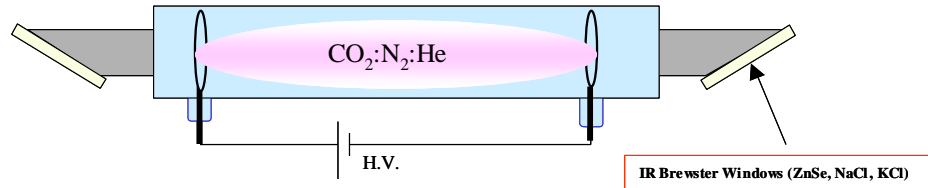
L=1 meter and P=25 torr

Need V=25 kV for optimum operation

11.4 Specific Types of CO₂ Lasers

High Power CW Operation

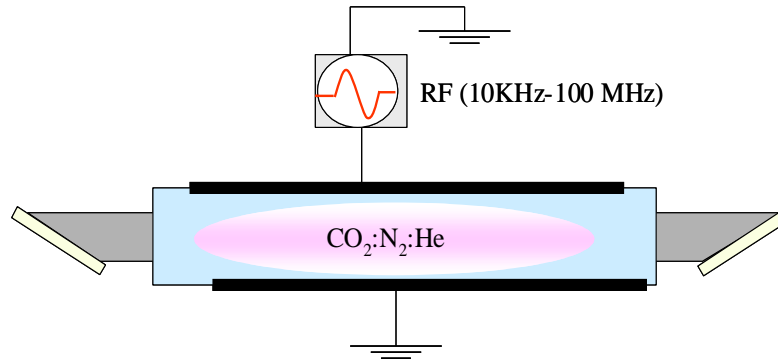
□ DC-Discharge



- Longitudinal discharge (High Voltage: 10-100 kV)
- Pressure: 10-100 torr
- Multistage discharge tubes can be used to produce kilowatts of output

power

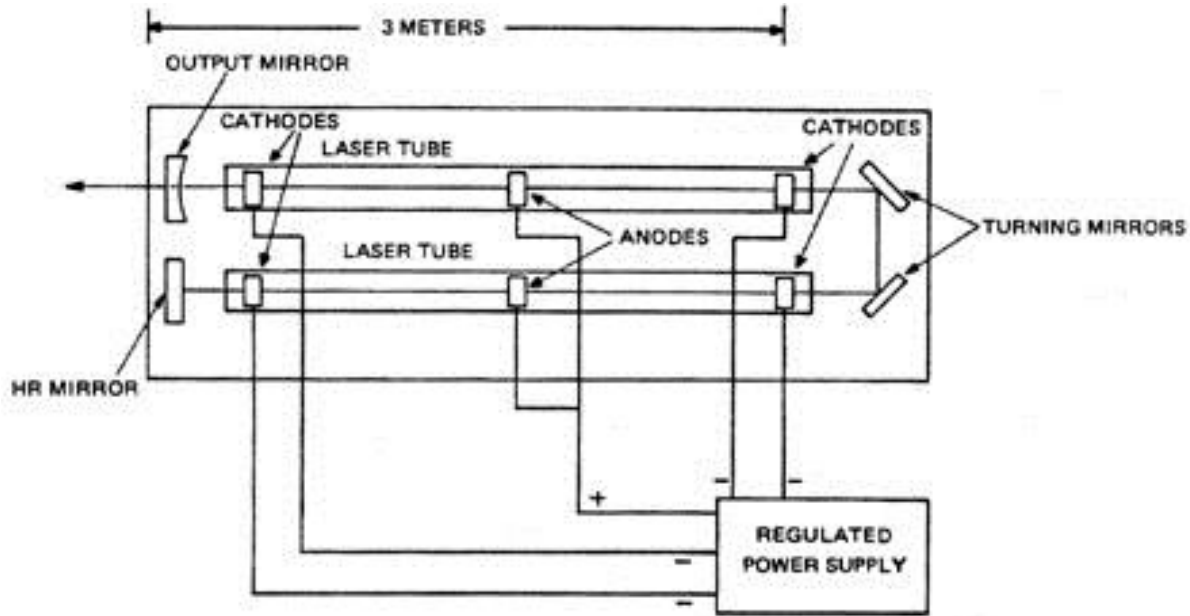
□ RF-Discharge



- In practice waveguides are used.
- High discharge stability, high pulsing frequency (up to 100 kHz)
- Expensive RF generator and requires EMI shielding

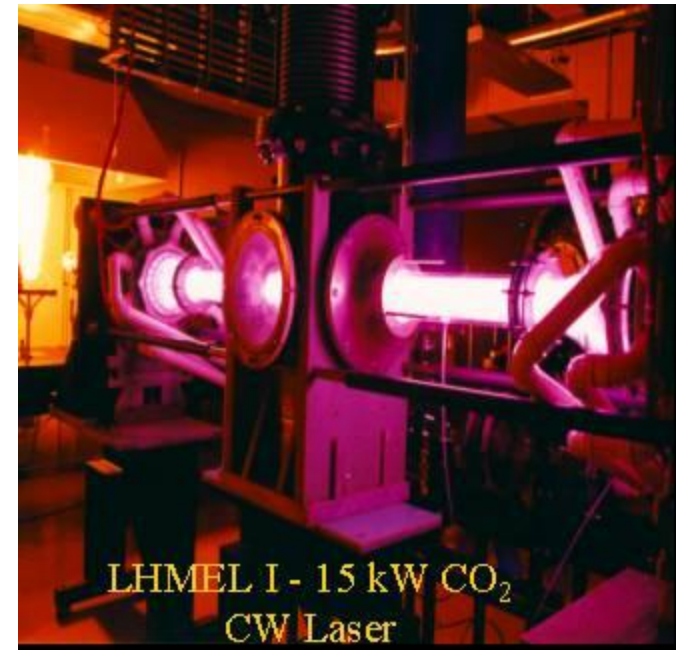
0.2 W/cm in a waveguide laser

Example: A 250 W CW CO₂ Laser

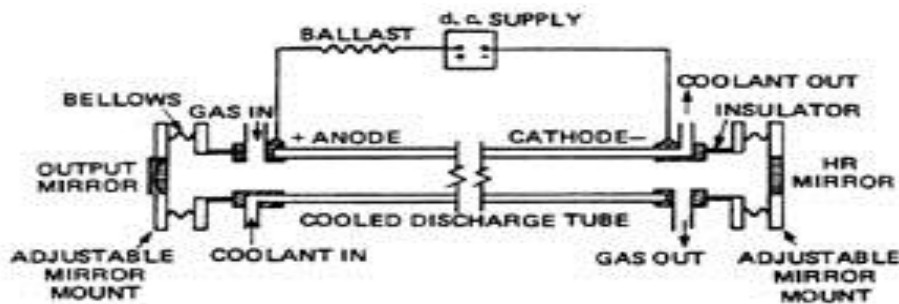


Operating Parameters of Commercial Class I CO₂ Lasers.

| Active Length meters | Output Power watts | Gas Mixture CO ₂ :N ₂ :He | Gas Flow Rate liters/min | Power/ Length W/m | Water Flow Rate liters/min |
|----------------------|--------------------|---|--------------------------|-------------------|----------------------------|
| 1 | 50 | 1:1.5:9.3 | 1.15 | 50 | 2 |
| 2 | 100 | 1:1.5:9.3 | 1.15 | 50 | 2 |
| 5 | 275 | 2:1.35:9.3 | 4.01 | 55 | 10 |
| 6 | 375 | 1:8:23 | 4.26 | 62.5 | 10 |
| 9 | 525 | 1:6.7:30 | 4.23 | 58.3 | 10 |
| 18 | 1000 | 1:2.35:17 | 14.35 | 55.6 | 15 |



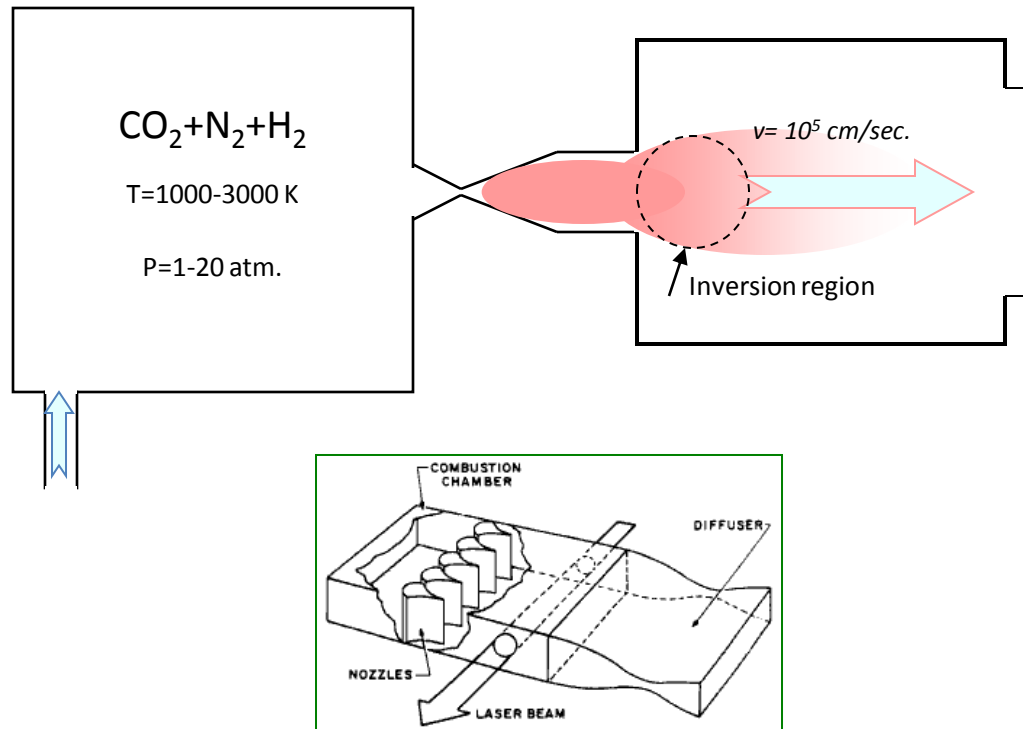
Electric Discharge Coaxial Laser (EDCL)



Gas-Dynamic Lasers

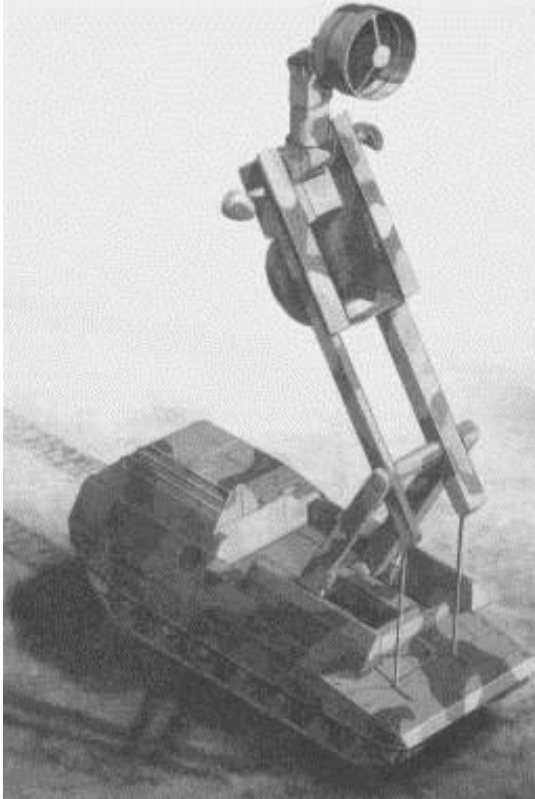
Basov & Oraevskii (1963)

Principle: Population inversion by rapid expansion (supersonic flow) of a super-heated gas



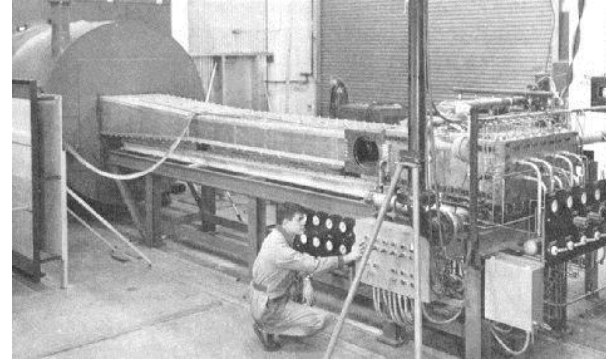
- cw powers up to 1 MW have been obtained from gas-dynamic CO_2 lasers !!

Gas-Dynamic Lasers

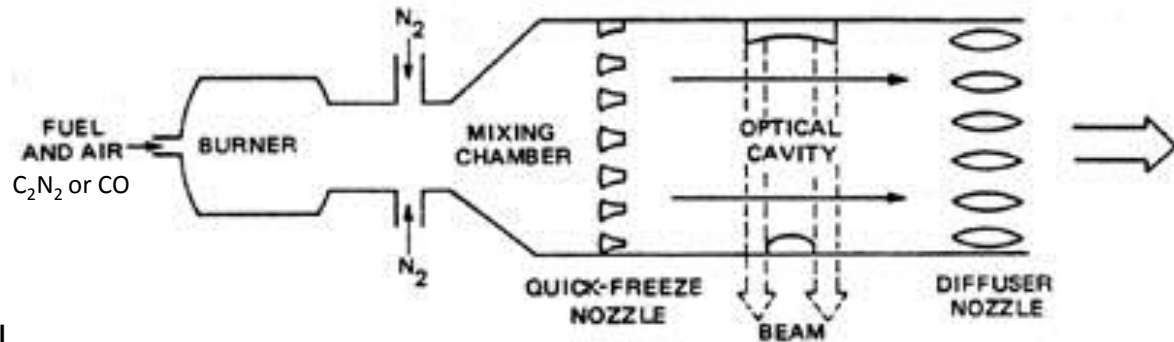


HELEX

High Energy Laser Experimental
Germany, 1970's

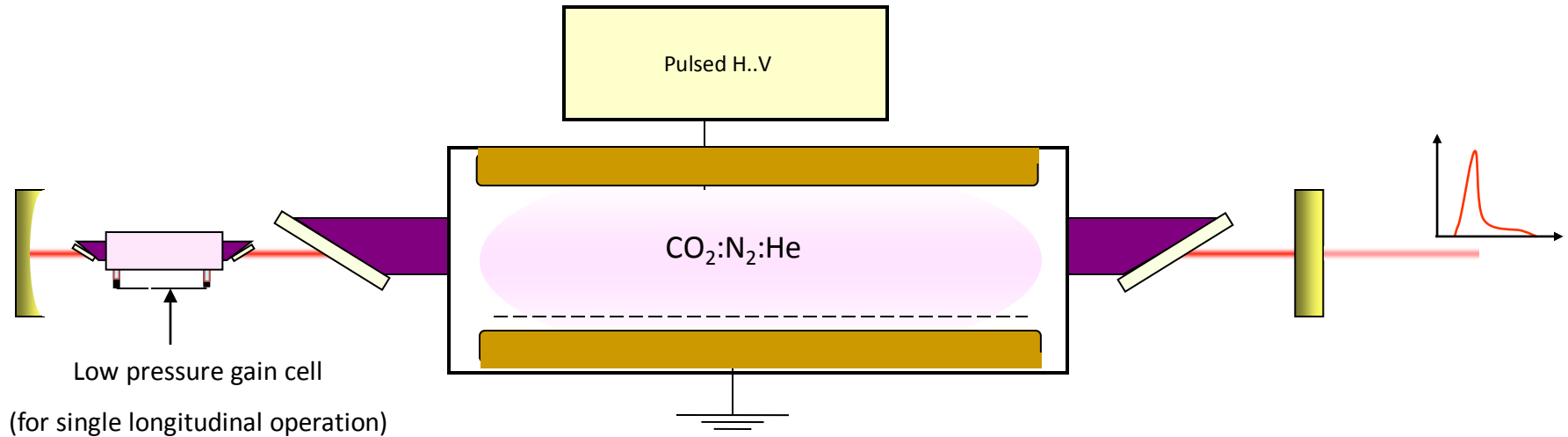


Large scale 135 Kilowatt gasdynamic laser at Avco Everett Research Lab.



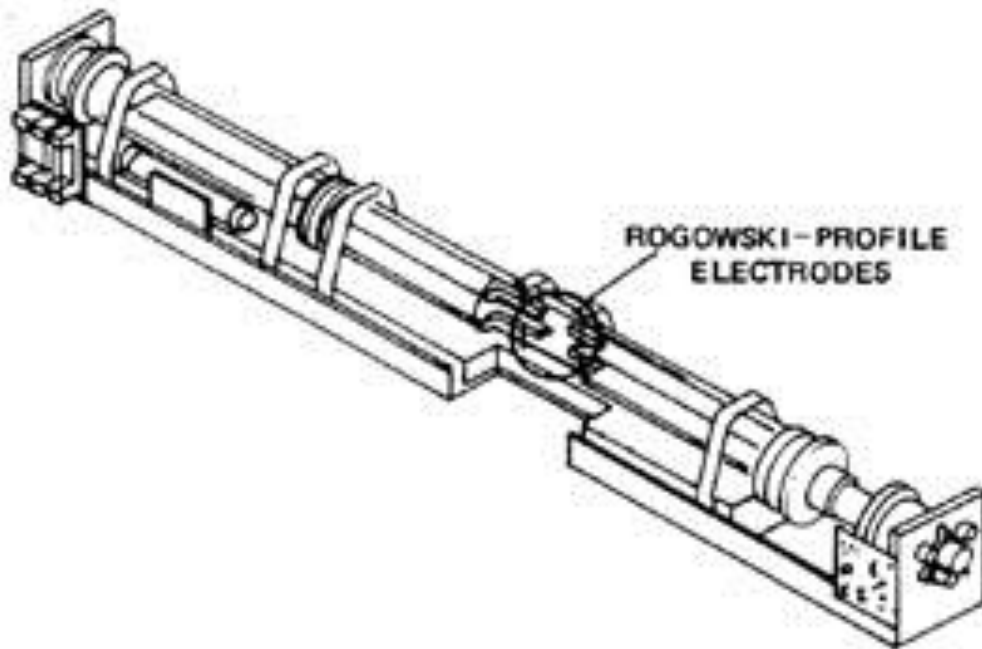
• Pulsed CO₂ Lasers

Most Common: Transversely Excited Atmospheric (TEA) CO₂ Lasers

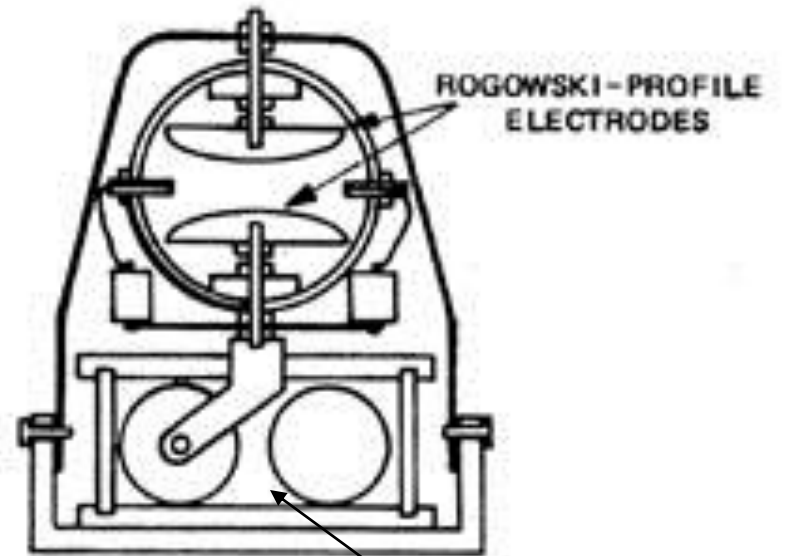


- Flowing or sealed systems
- Pulsewidths from 50 ns to 300 ns
- Repetition rates: 1 Hz. to 1 kHz.
- Pulse energy: 50 mJ to 10 J (amplified)

Example



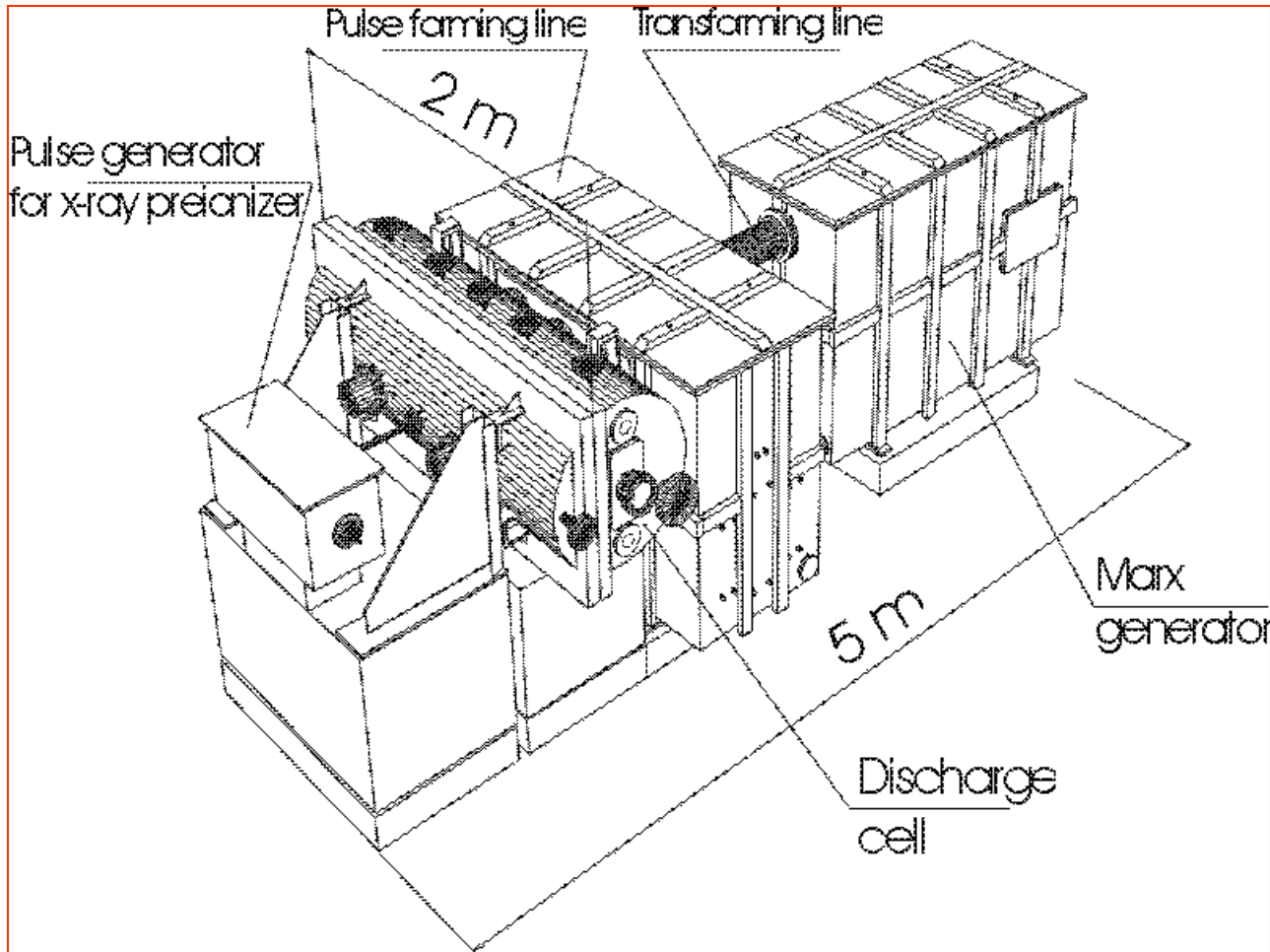
a. Perspective view

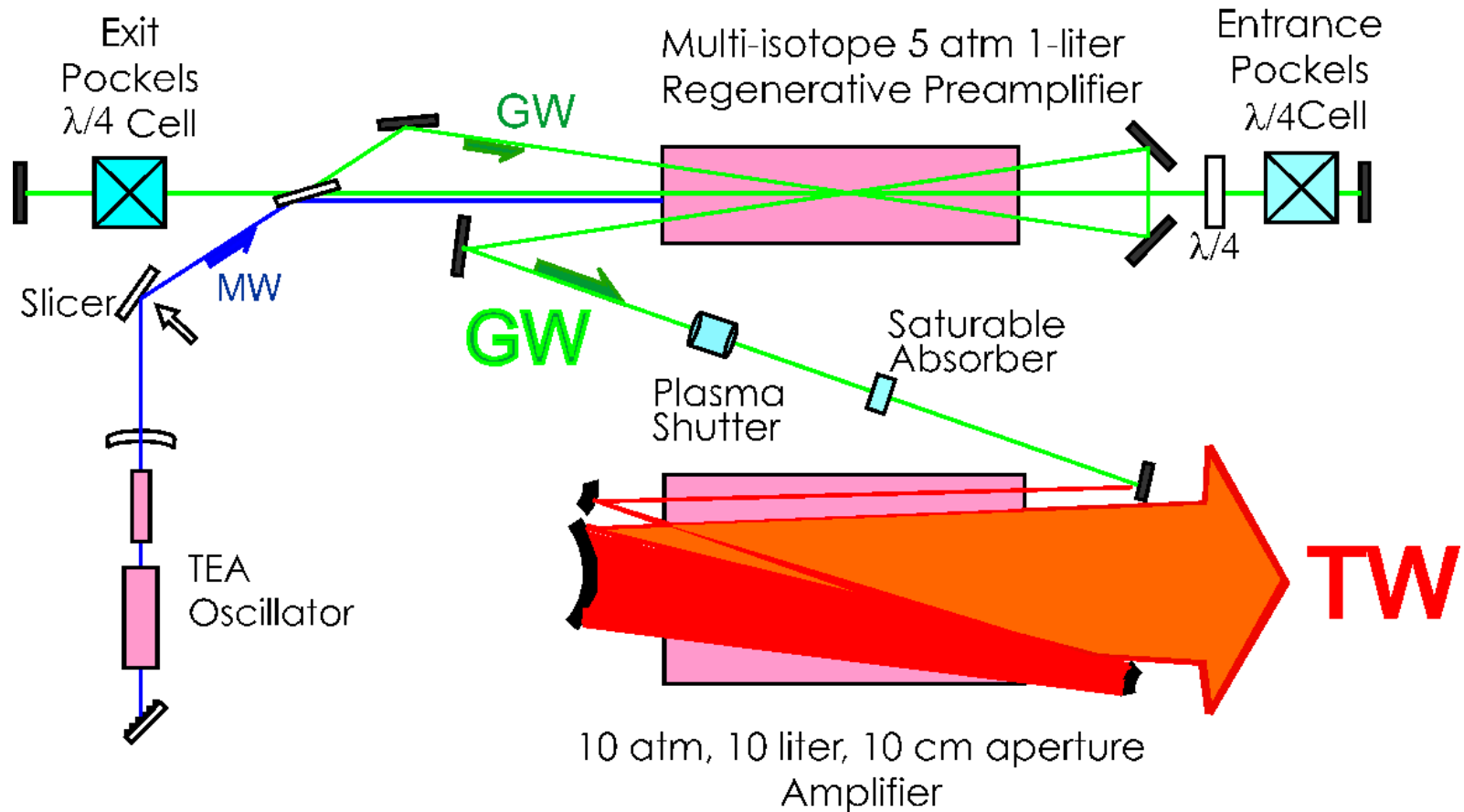


b. Cross-sectional view

Capacitor bank

Terra Watts Pulsed CO₂ Lasers



Picosecond TW CO₂ Laser at BNL

Excimer lasers:

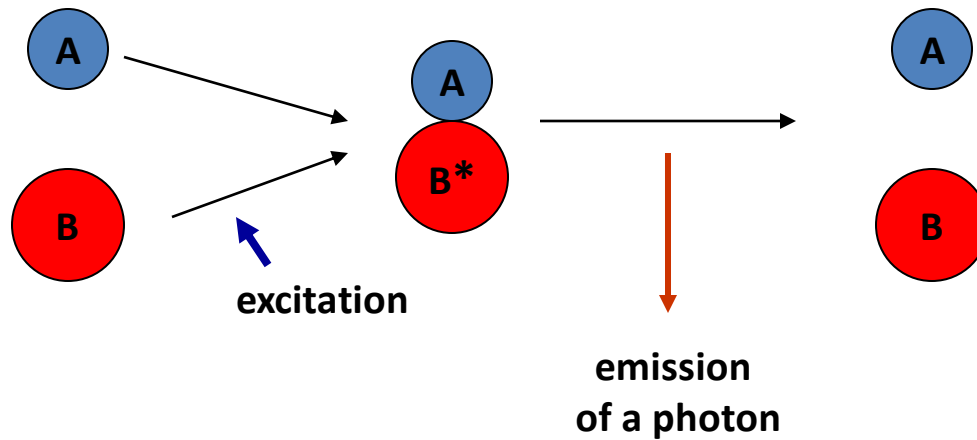
applications in lithography and eye surgery

molecules exist only in the excited state

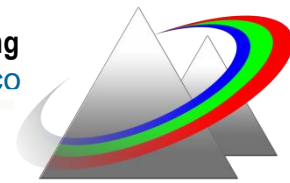
XeCl 308 nm

KrF 248 nm

ArF 193 nm

F₂ 156 nm

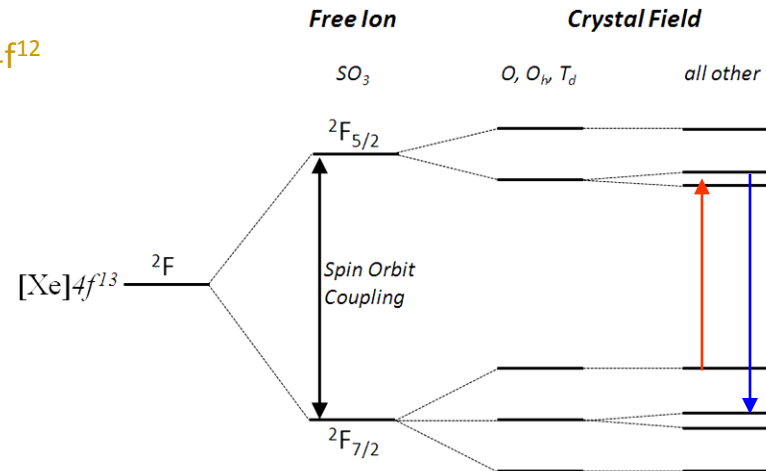
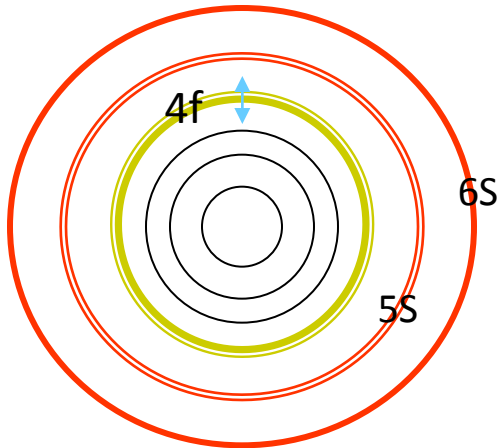
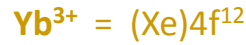
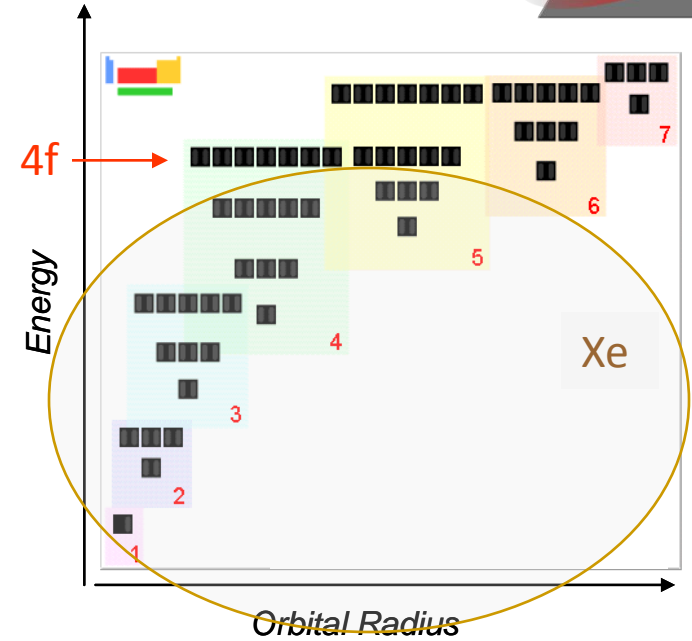
The 4f-4f transitions in Rare-Earths Ions:



| | | | | | | | | | | | | | | | | | |
|----------------------------------|---------------------------------|------------------------------------|-------------------------------------|----------------------------------|----------------------------------|---------------------------------|---------------------------------|----------------------------------|---------------------------------|--------------------------------|----------------------------------|----------------------------------|-----------------------------------|---------------------------------|----------------------------------|---------------------------------|------------------------------|
| 1A 1 | | | | | | | | | | | | | | | | | 8A 18 |
| 1 H Hydrogen 1.00794 | | | | | | | | | | | | | | | | | 2 He Helium 4.00260 |
| 3 Li Lithium 6.941 | 4 Be Beryllium 9.01218 | | | | | | | | | | | 5 B Boron 10.811 | 6 C Carbon 12.011 | 7 N Nitrogen 14.0067 | 8 O Oxygen 15.9994 | 9 F Fluorine 18.998403 | 10 Ne Neon 20.1797 |
| 11 Na Sodium 22.98977 | 12 Mg Magnesium 24.305 | 3B 3 | 4B 4 | 5B 5 | 6B 6 | 7B 7 | 8B 8 9 10 | | 1B 11 | 2B 12 | 13 Al Aluminum 26.98154 | 14 Si Silicon 28.0855 | 15 P Phosphorus 30.97376 | 16 S Sulfur 32.06 | 17 Cl Chlorine 35.4527 | 18 Ar Argon 39.948 | |
| 19 K Potassium 39.0983 | 20 Ca Calcium 40.078 | 21 Sc Scandium 44.9559 | 22 Ti Titanium 47.88 | 23 V Vanadium 50.9415 | 24 Cr Chromium 51.9961 | 25 Mn Manganese 54.938 | 26 Fe Iron 55.847 | 27 Co Cobalt 58.9332 | 28 Ni Nickel 58.6934 | 29 Cu Copper 63.546 | 30 Zn Zinc 65.39 | 31 Ga Gallium 69.723 | 32 Ge Germanium 72.61 | 33 As Arsenic 74.9216 | 34 Se Selenium 78.96 | 35 Br Bromine 79.904 | 36 Kr Krypton 83.80 |
| 37 Rb Rubidium 85.4678 | 38 Sr Strontium 87.62 | 39 Y Yttrium 88.9059 | 40 Zr Zirconium 91.224 | 41 Nb Niobium 92.9064 | 42 Mo Molybdenum 95.94 | 43 Tc Technetium (98) | 44 Ru Ruthenium 101.07 | 45 Rh Rhodium 102.9055 | 46 Pd Palladium 106.42 | 47 Ag Silver 107.8682 | 48 Cd Cadmium 112.411 | 49 In Indium 114.82 | 50 Sn Tin 118.710 | 51 Sb Antimony 121.757 | 52 Te Tellurium 127.603 | 53 I Iodine 126.9045 | 54 Xe Xenon 131.29 |
| 55 Cs Cesium 132.9054 | 56 Ba Barium 137.327 | *57 La Lanthanum 138.9055 | 72 Hf Hafnium 178.49 | 73 Ta Tantalum 180.9479 | 74 W Tungsten 183.85 | 75 Re Rhenium 186.207 | 76 Os Osmium 190.2 | 77 Ir Iridium 192.22 | 78 Pt Platinum 195.08 | 79 Au Gold 196.9665 | 80 Hg Mercury 200.59 | 81 Tl Thallium 204.3833 | 82 Pb Lead 207.2 | 83 Bi Bismuth 208.9804 | 84 Po Polonium (209) | 85 At Astatine (210) | 86 Rn Radon (222) |
| 87 Fr Francium 226.0254 | 88 Ra Radium 226.0254 | †89 Ac Actinium 227.0278 | 104 Rf Rutherfordium (261) | 105 Db Dubnium (262) | 106 Sg Seaborgium (263) | 107 Bh Bohrium (262) | 108 Hs Hassium (265) | 109 Mt Meitnerium (268) | 110 | 111 | 112 | | | | | | |

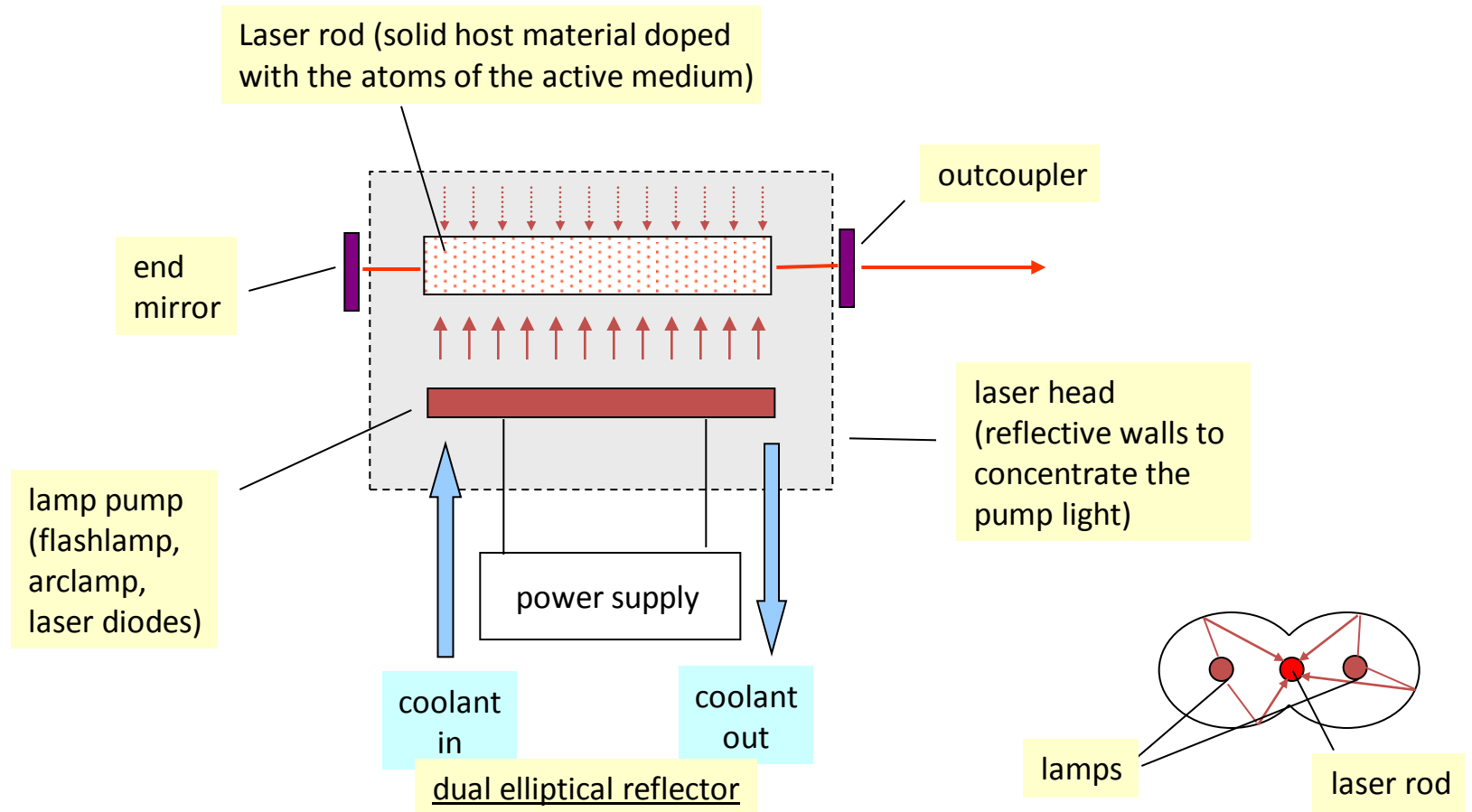
| | | | | | | | | | | | | | | |
|--------------------|---------------------------------|--------------------------------------|---------------------------------|----------------------------------|--------------------------------|---------------------------------|----------------------------------|---------------------------------|----------------------------------|----------------------------------|-------------------------------|-----------------------------------|----------------------------------|----------------------------------|
| *Lanthanide Series | 58 Ce Cerium 140.115 | 59 Pr Praseodymium 140.9077 | 60 Nd Neodymium 144.24 | 61 Pm Promethium (145) | 62 Sm Samarium 150.36 | 63 Eu Europium 151.965 | 64 Gd Gadolinium 157.25 | 65 Tb Terbium 158.9254 | 66 Dy Dysprosium 162.50 | 67 Ho Holmium 164.9303 | 68 Er Erbium 167.26 | 69 Tm Thulium 168.9342 | 70 Yb Ytterbium 173.054 | 71 Lu Lutetium 174.967 |
| † Actinide Series | 90 Th Thorium 232.0381 | 91 Pa Protactinium 231.0369 | 92 U Uranium 238.0289 | 93 Np Neptunium 237.048 | 94 Pu Plutonium (244) | 95 Am Americium (243) | 96 Cm Curium (247) | 97 Bk Berkelium (247) | 98 Cf Californium (251) | 99 Es Einsteinium (252) | 100 Fm Fermium (257) | 101 Md Mendelevium (258) | 102 No Nobelium (259) | 103 Lr Lawrencium (260) |

Central Washington University © 1998

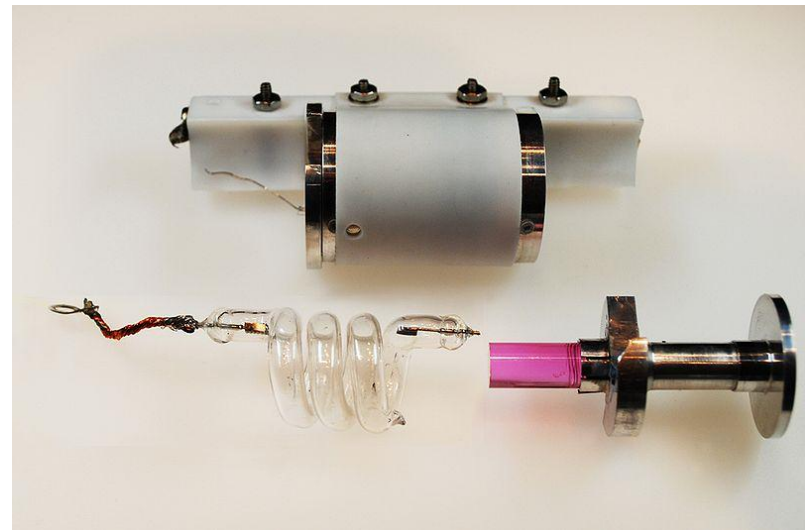
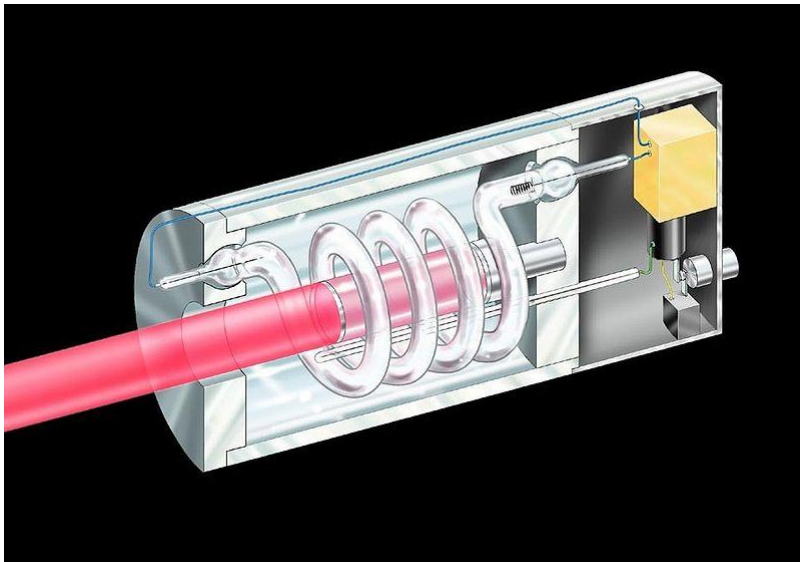


| | | He-Ne | Nd:YAG | Ti:Al ₂ O ₃ |
|--|--|--|---|---|
| | λ | 633 nm | 1064 nm | 700 - 1000 nm |
| SSL: much smaller cross sections | σ | $3 \times 10^{-13} \text{ cm}^{-2}$ | $4 \times 10^{-19} \text{ cm}^{-2}$ | $4 \times 10^{-19} \text{ cm}^{-2}$ |
| SSL: much longer lifetimes | τ_2 / τ_1 | 60 ns / 10 ns | 250 μs / ~ 30 ns | 3.2 μs / fast |
| SSL: much higher saturation intens. | I_{sat} | 2 W/cm ² | 2 kW/cm ² | 200 kW/cm ² |
| SSL: much smaller quantum defect | $\frac{E_{\text{laser}}}{E_{\text{pump}}}$ | $\frac{2 \text{ eV}}{20 \text{ eV}}$ | $\frac{1.2 \text{ eV}}{1.5 \text{ eV}}$ | $\frac{1.6 \text{ eV}}{2.3 \text{ eV}}$ |
| SSL: much(!) wider gain spectra | $\Delta\omega$ | $2\pi \times 1.5 \text{ GHz}$ | $2\pi \times 200 \text{ GHz}$ | $2\pi \times 100 \text{ THz}$ |
| SSL: atom lifetime \gg cavity lifetime | τ_c | 200 ns | 5 – 100 ns | 5 – 100 ns |
| §SSL: much higher number densities | N | 1 torr $3 \times 10^{16} \text{ cm}^{-3}$ | 1 at.%, $1.4 \times 10^{20} \text{ cm}^{-3}$ | 0.5 wt.%, $1.7 \times 10^{20} \text{ cm}^{-3}$ |

10.2 Layout of a solid-state laser



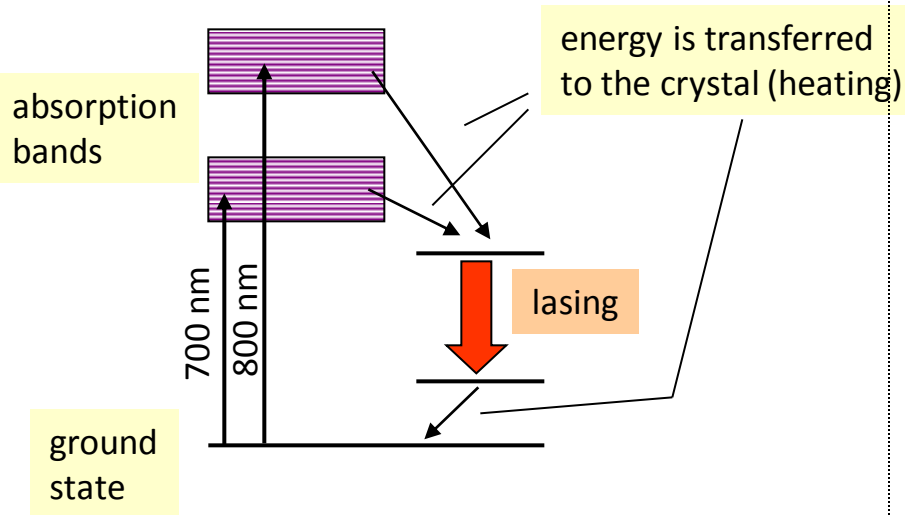
Maiman's Ruby Laser



10.3 Nd:YAG laser

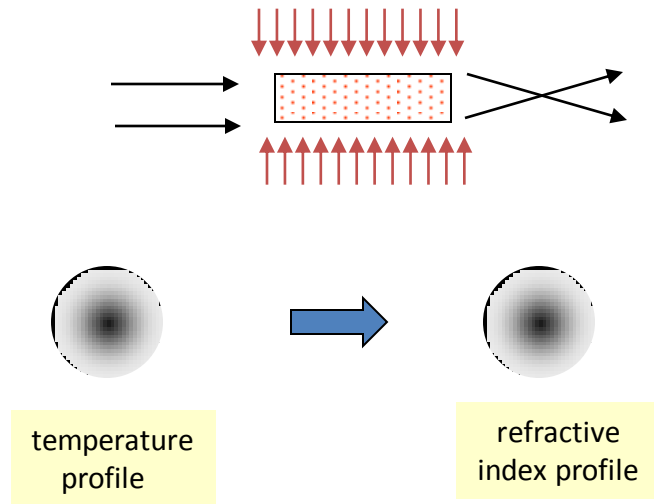
The most common solid-state laser is based on Nd atoms as dopands.

Energy diagram of Nd:



By changing the host material the laser wavelength and the thermal properties can be changed.

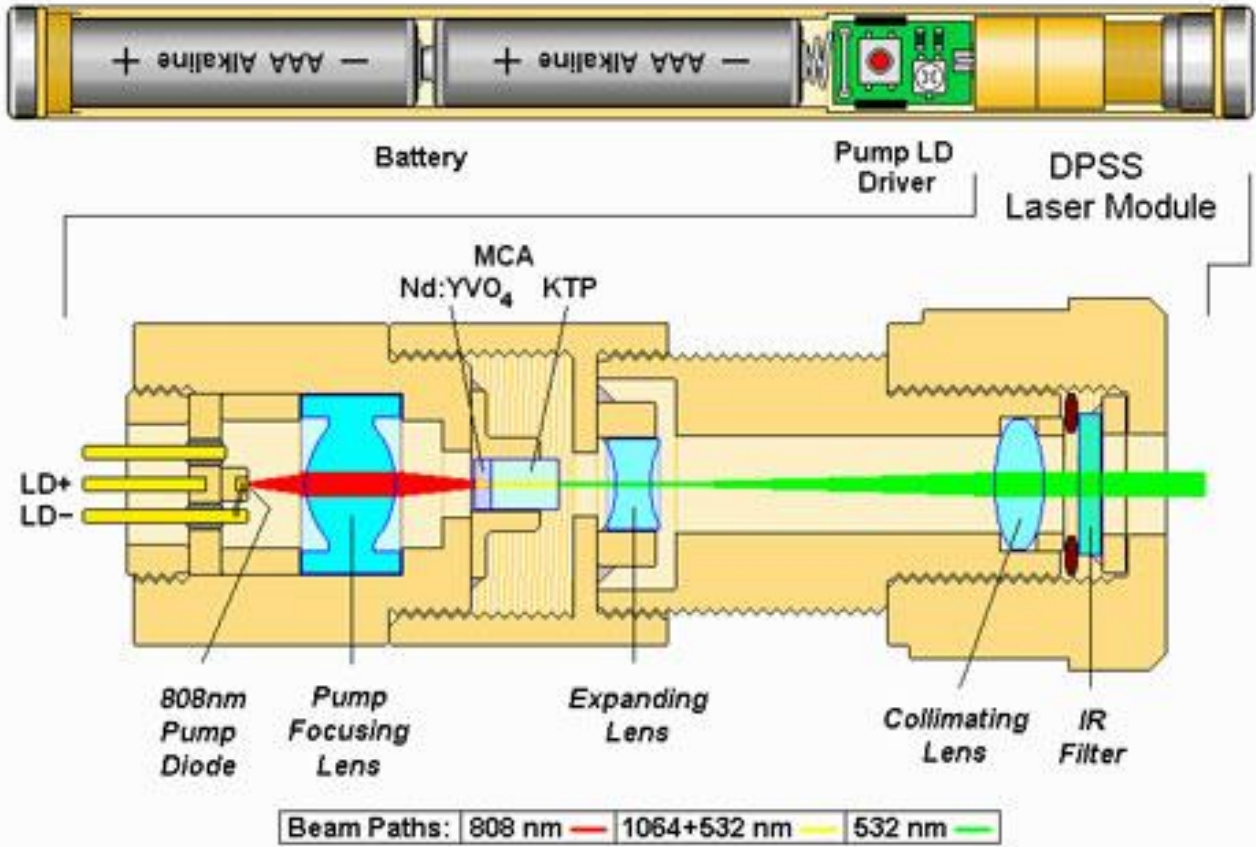
thermal effects:



Output (Nd:YAG)

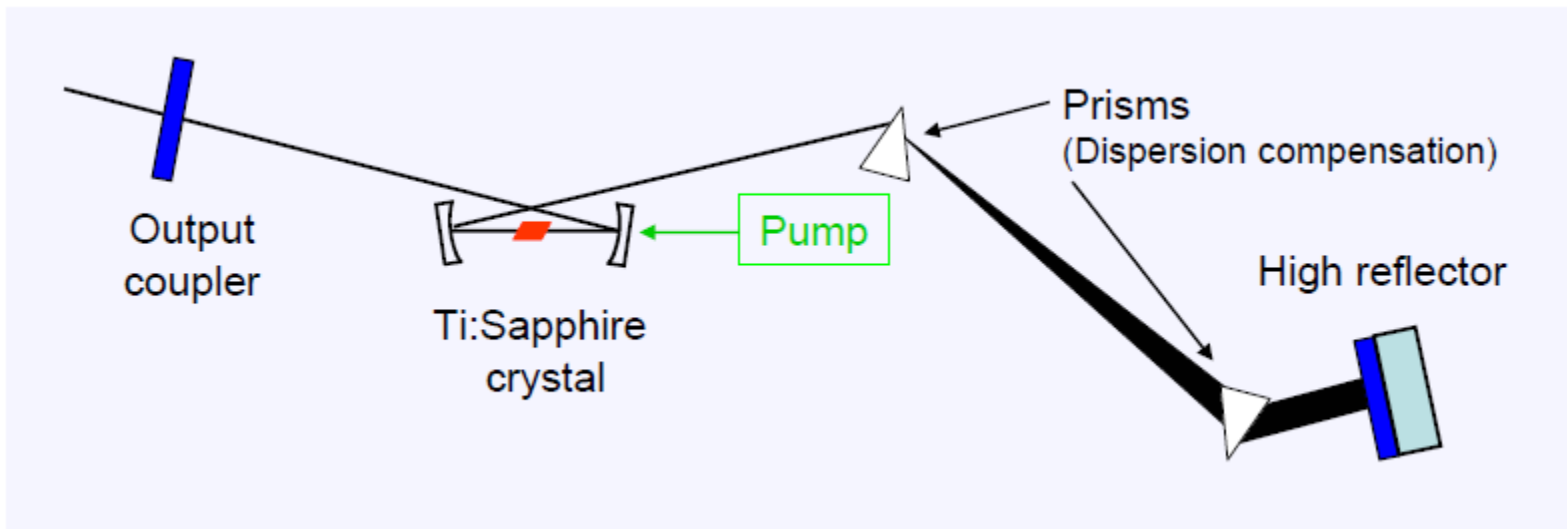
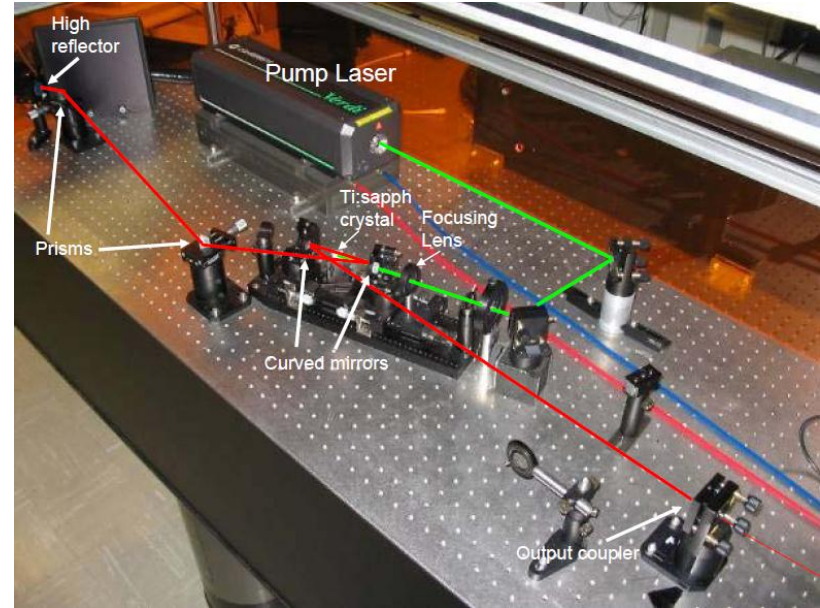
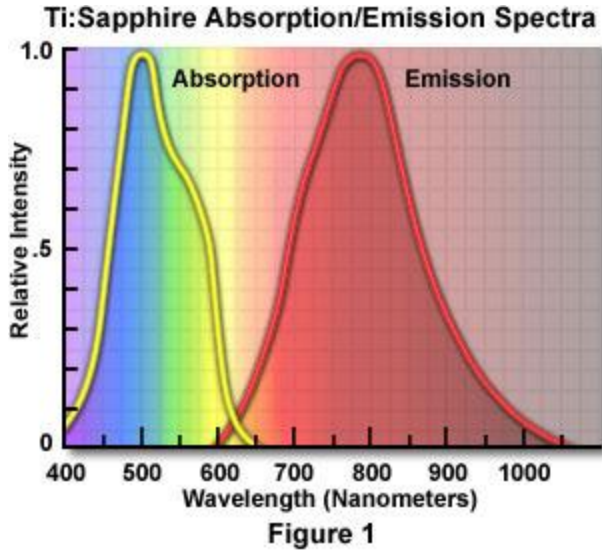
cw: ≤ 1000 W
pulsed: pulse energy ≤ 1 Joule
Q-switched - 10 ns pulse duration
modelocked - 100 ps pulse duration

Green Laser Pointer: a frequency doubled diode-pumped Nd:YVO₄ Laser!!



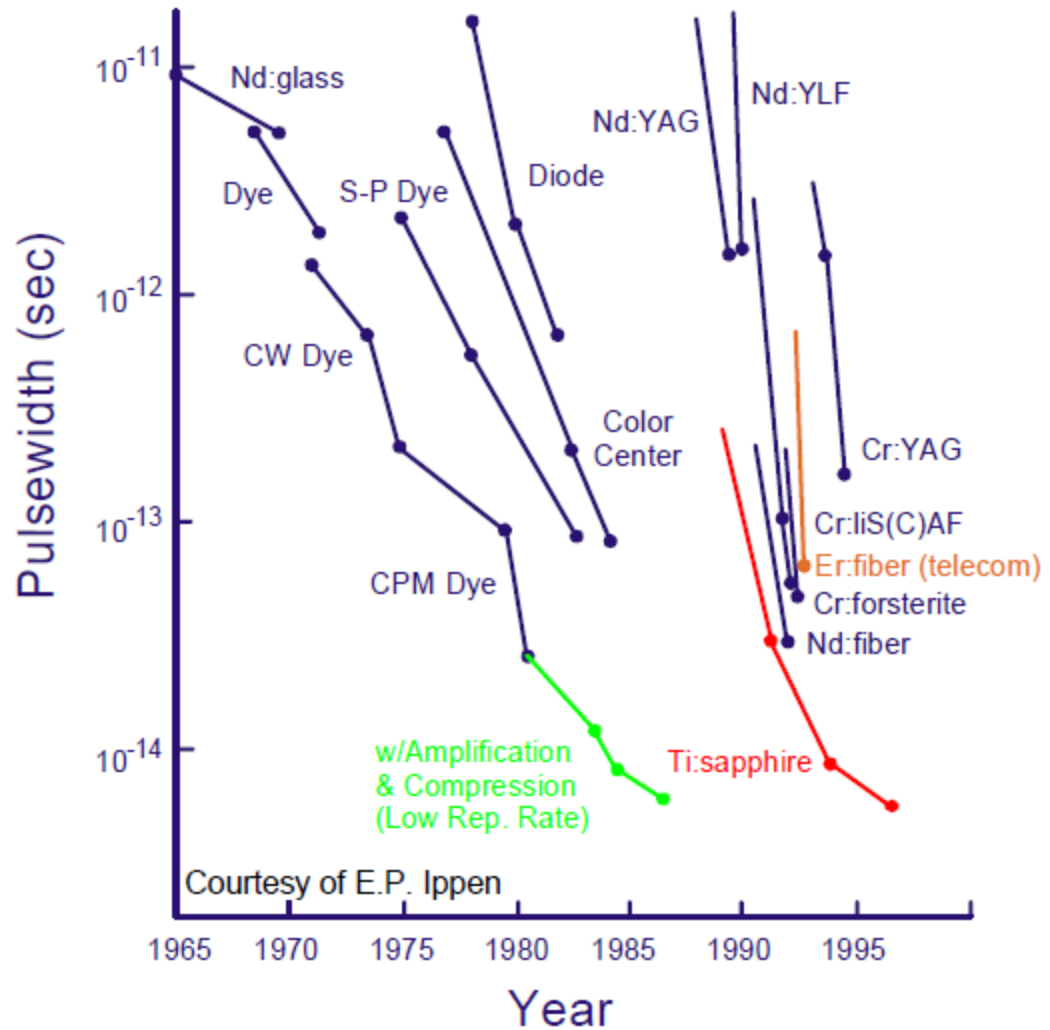
Titanium doped sapphire (Ti:Al₂O₃) laser

The jewel of ultrafast lasers!!



Historical Progress in Ultrashort Pulses

ADVANCES IN SHORT PULSE GENERATION

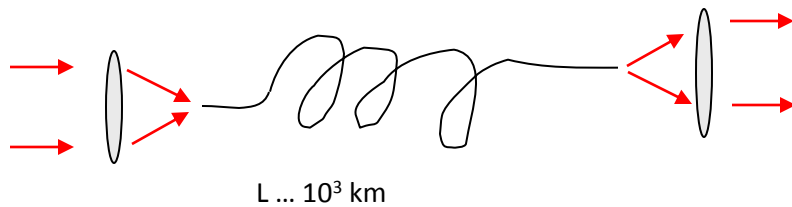
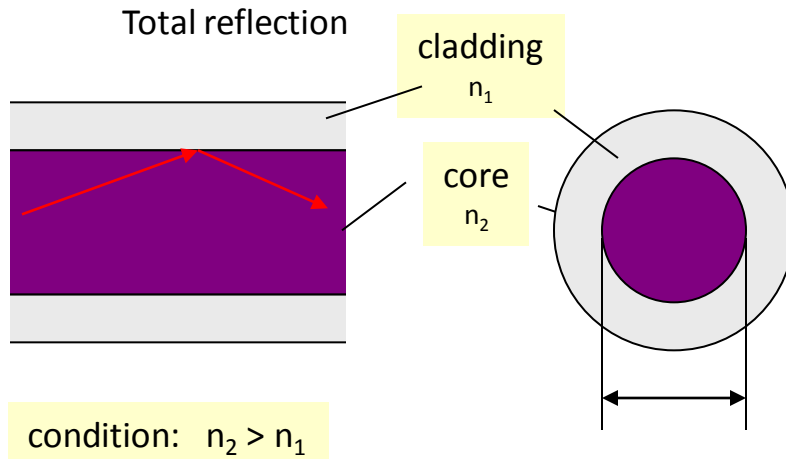


Courtesy of E.P. Ippen

11. Fiber lasers

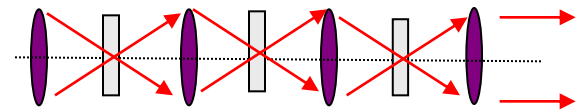
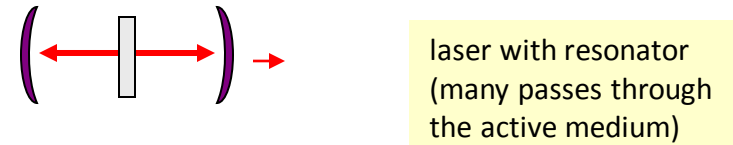
11.1 Introduction

Optical fiber

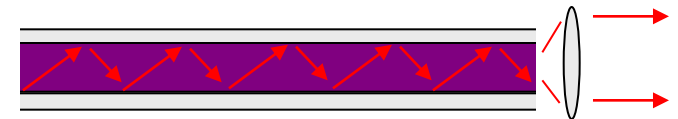


Light can be guided (confined) in the core of optical fibers over great distances. This allows for large interaction lengths of light with an active medium that is doped into the fiber core.

Realizing large gain

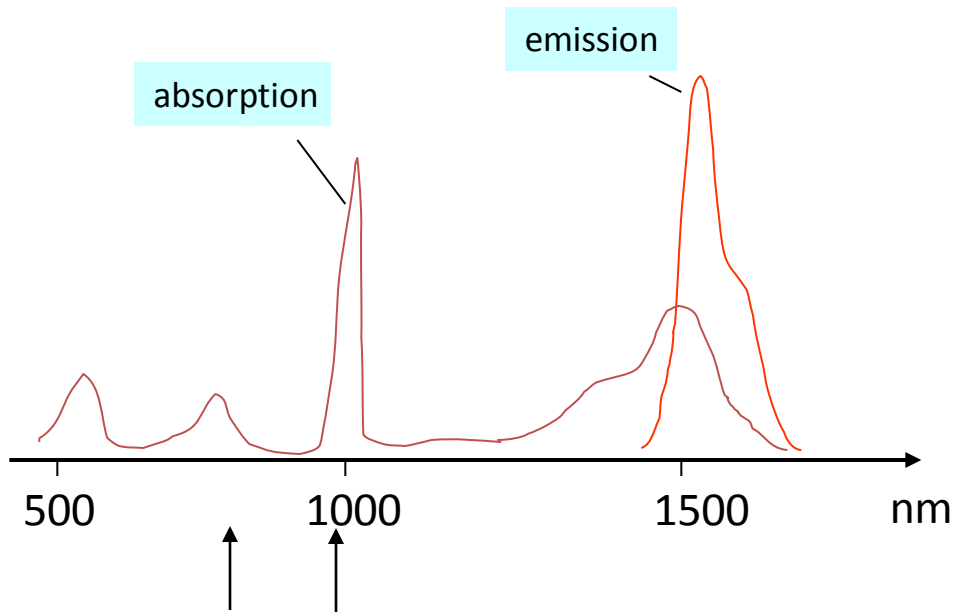


lens duct (unfolded resonator)



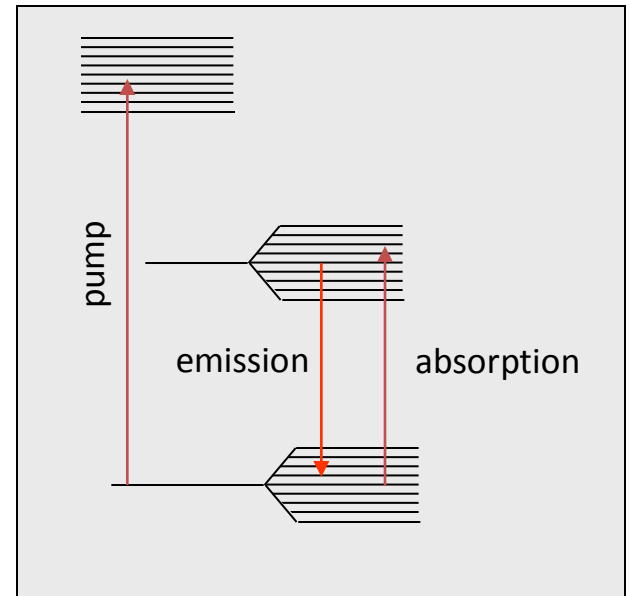
fiber laser

11.2 Example: erbium-doped glass fibers

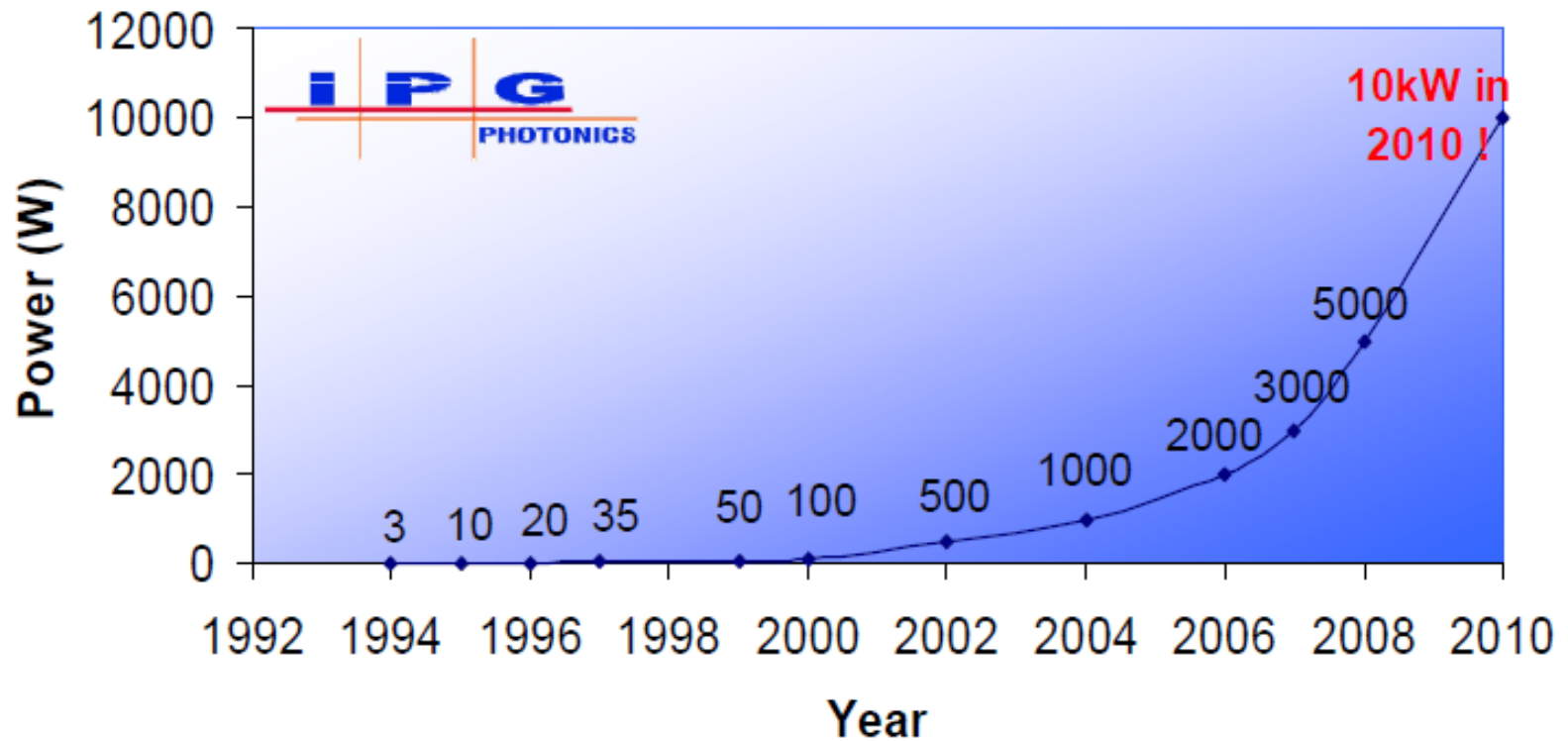


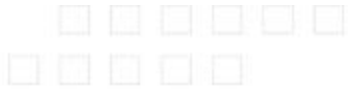
pumping with diode lasers is possible

The wavelength of about 1550 nm is particularly interesting for applications in telecommunication.



Growth of Yb:HPFL SM (near diffraction limited)





IPG Fiber Lasers

IPG Fiber Lasers: The technology



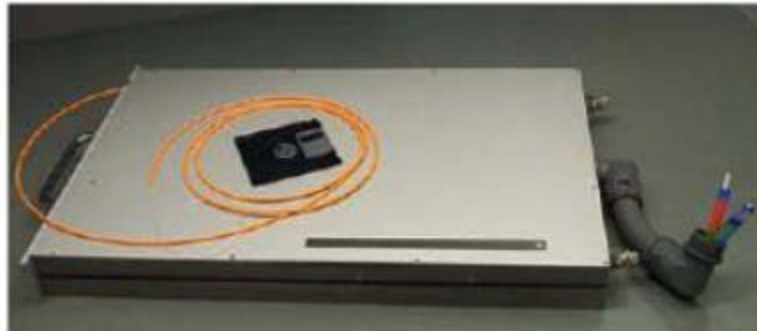
A single module can supply:

- 250, 400, 800, 1000+ W of laser power
- Wavelength of 1070nm (NIR)
- One 7 or 15 μm fiber core
- 0.34-0.41mm*mrad beam divergence

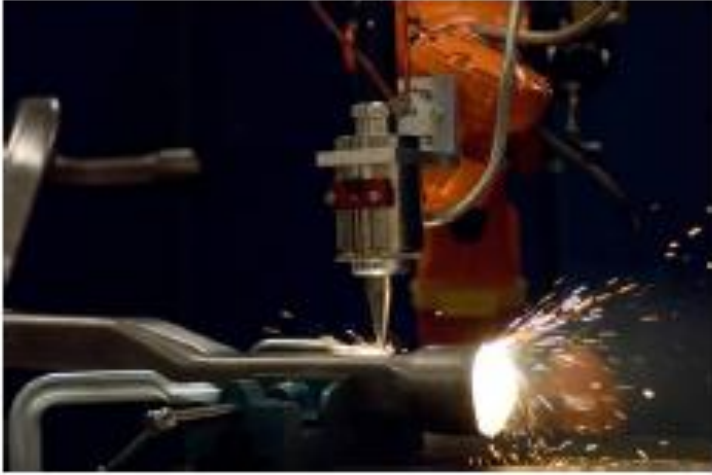
T x H x D = 60 x 33 x 4.7 cm

Efficiency (DC) > 35%

Building blocks (modules) for HPFLs

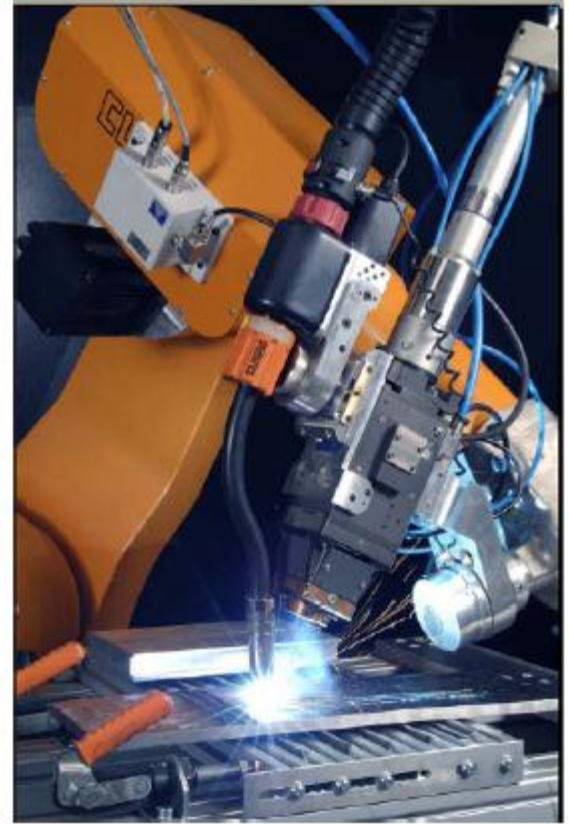


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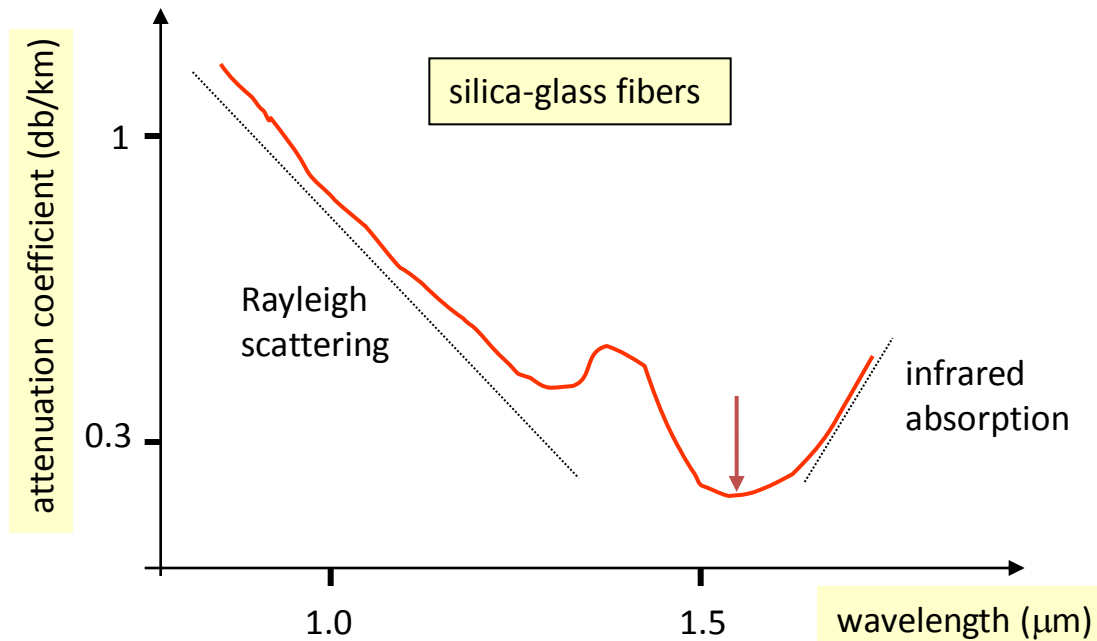


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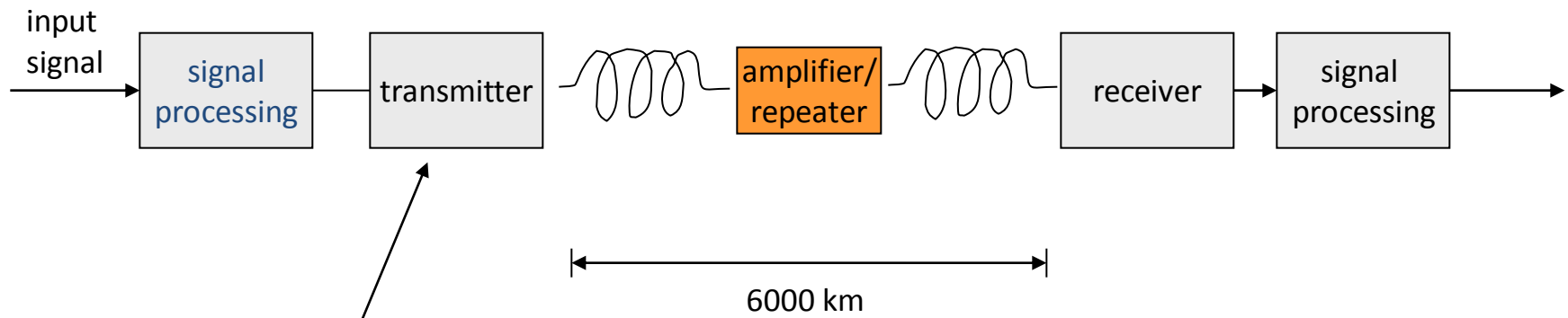
11.3 Fiber-optic Communications

Why?

The carrier frequency of light ($\sim 10^{14}$ Hz) and subsequently the transmitted bandwidth is much larger than what can be achieved by electronics.



Fiber transmission line



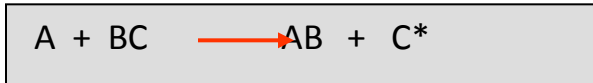
InGaAsP diode laser

- transatlantic US - UK
- 560 Mb/s per fiber pair
- 80000 simultaneous voice channels
- repeaters 100 km apart

12. Chemical Lasers

12.1 Introduction

- population inversion is produced by a chemical reaction



- electrical power supply is not needed
- airborne lasers
- first chemical laser: 1964

chemical reaction:

- exothermic
- generation rate must be large enough to overcome spontaneous emission and collisional relaxation

Examples:

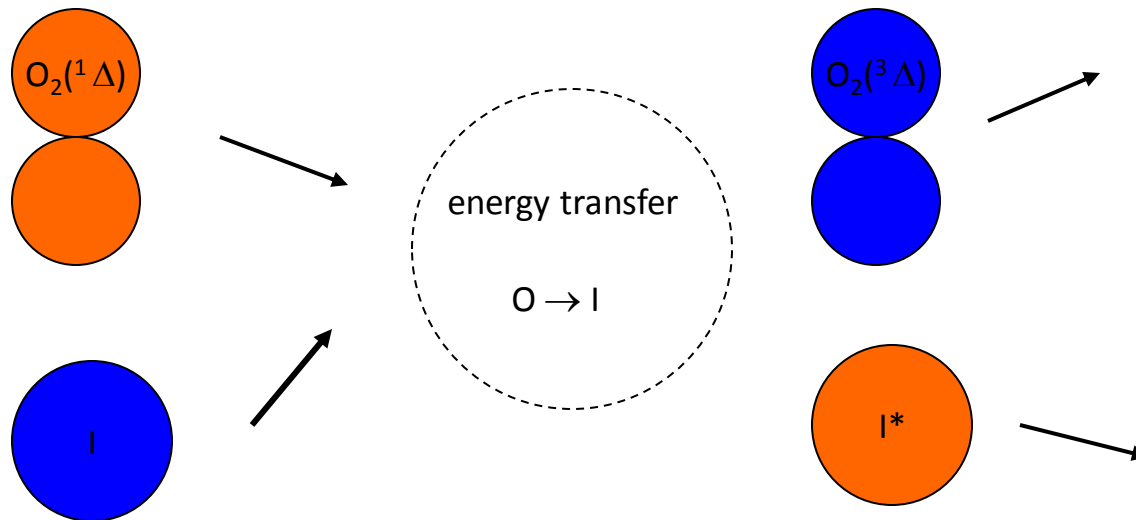
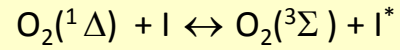
| reaction | active medium | wavelength |
|-----------------------------------|---------------|-------------------------|
| $F + D_2 \rightarrow DF^* + D$ | DF | 3.5 - 4.1 μm |
| $Cl + HI \rightarrow HCl^* + I$ | HCl | 3.5 - 4.1 μm |
| $H + Br_2 \rightarrow HBr^* + Br$ | HBr | 4.0 - 4.7 μm |
| $F + H_2 \rightarrow HF^* + H$ | HF | 3.5 - 4.1 μm |
| $I + O_2^* \rightarrow I^* + O_2$ | I | 1.31 μm |

molecules in an excited vibrational state

atoms in an excited electronic state

12.2 The chemical oxygen-iodine laser

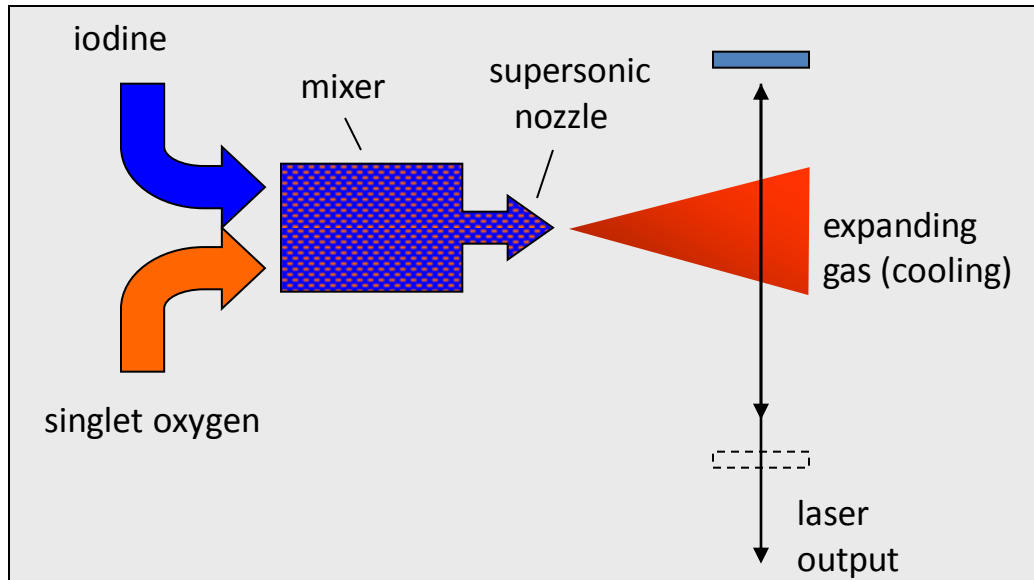
chemical reaction:



steps:

1. generation of singlet oxygen $\text{Cl}_2 + \text{H}_2\text{O}_2 + 2\text{NaOH} \rightarrow \text{O}_2(^1\Delta) + 2\text{H}_2\text{O} + 2\text{NaCl}$
2. production of excited iodine $\text{O}_2(^1\Delta) + \text{I} \leftrightarrow \text{O}_2(^3\Sigma) + \text{I}^*$
3. lasing of excited iodine

schematic diagram of a chemical iodine laser



parameters

- MW output power
- wavelength 1.315 micron
- pulsed and cw

atmospheric absorption

